

ACS Lab Safety Education Resources

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Introductions

Ralph

- Environmental Safety Manager and Chemical Hygiene Officer at Keene State College
- Previously at UVM and Cornell
- Chair of the ACS Committee on Chemical Safety
- Membership Chair of the ACS Division of Chemical Health and Safety



You

- Name
- Institution
- Job Title
- Biggest Chemical Safety Challenge

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Goals of the Workshop

1. Review lab chemical safety tools developed by the ACS
2. Use these tools in specific examples and place them in the context of an academic laboratory chemical safety program
3. Discuss the strengths and weaknesses of these tools from the EHS point of view and generate ideas for future ACS tools

Goals

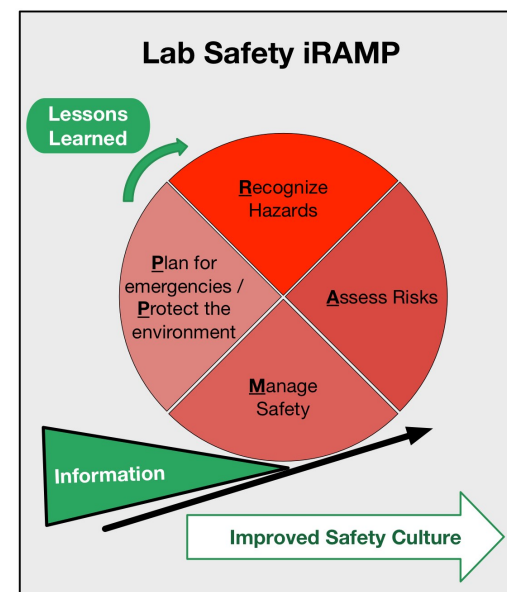


Moving Lab Safety into the 21st Century

Lab Safety involves both Technical and Cultural Skills

20th Century:
Selecting Controls
Based on Rules, guided by
Chemical Intuition

21st Century:
a Safety System based
on documented
Risk Assessment



Chemical Safety in the 20th Century



In 1964, the Journal of Chemical Education published an article by Dr. Livingston, entitled ***Safety Considerations in Research Proposals***,

- The article is a good summary of the research safety challenges that still apply today.
- However he states: ***“Legal requirements... are outside the competence of our committee... Certainly if humanitarian and ethical requirements are met, there are not likely to be any issues that will require legal action.”***
- When events of the 1980’s pushed ***Chemical Safety and Hygiene*** to become ***Environmental Health and Safety***, this “gentleman’s club” approach to lab safety culture became outdated



H.K. Livingston, first CCS chair in 1963, newly moved to Wayne State University after 13 years at DuPont

“Working Safely at the Frontiers of Science”

In a 1999 JCHAS interview with Dr. Seaborg remembered:

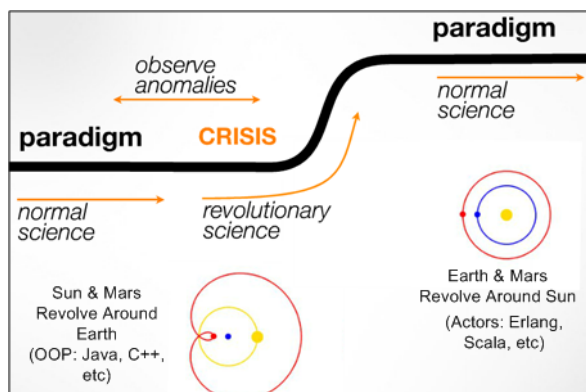
- ...it suddenly occurred to me that the ... **health physicists hadn't given any attention to the danger from alpha-particle emitters like plutonium.** All of the precautions... were for gamma radiation.
- “In view of the problems that had occurred in the late 1910's... with the **radium dial painters**, I realized that the ingestion of just a little bit of plutonium would be a greater danger than radiation from gamma emitters.
- **“So I got in touch with the medical authorities and called the danger to their attention.** This led to a recognition of the problem and a renovation of the entire laboratory to include additional hood space and air monitoring.”

Dr. Seaborg's experience with the changing nature of “safety” as science advanced led to him supporting establishing DCHAS in 1979 before the ACS Council, despite DAC opposition. The motion to approve the Division carried.



Glenn Seaborg,
ACS President, 1976;
patent holder on
americium and curium

The Paradigm Shift: 21st Century Safety Culture includes Community Safety as well as Personal Safety



Paradigm 1: rules based safety
Crisis: CSB report, *Safe Science*
Paradigm 2: risk based safety

- **Community safety** requires adding *transparency, transferability, scalability* and *sustainability* to the rules-based hazard management process. These values are the basis of scientific culture as well.
- The cultural stress resulting from this change is seen in the 2018 ACS Strategic Plan statement: *“Despite increasing awareness of the importance of having an active safety culture in the workplace, some practitioners see safety as interfering with success.”*

Safety within the ACS Strategic Plan



STRATEGIC PLAN for 2018 and Beyond

Vision

Improving people's lives through the transforming power of chemistry

Mission

Advancing the broader chemistry enterprise and its practitioners for the benefit of Earth and its people

ACS Safety Strategic Goals

1. Provide Chemical Safety Information Solutions
2. Empower Members and Member Communities to be Safety Leaders
3. Support Excellence in Safety Education
4. Communicate the Value of Safety to the Chemical Enterprise

Empowering ACS Members to Meet These Challenges



Technical Skills

- Understanding GHS
- Using the RAMP paradigm
- Effectively participating in peer safety reviews
- Maintaining situational awareness during chemical processes



Cultural Skills

Within the project team:

- Asking Effective Questions (*empowerment*) and
- Anticipating Others' Challenges and Sharing Lessons Learned (*leadership*)

Outside the project team:

- Understanding Legal Expectations
- Participating in two-way Risk Communication

Personal development:

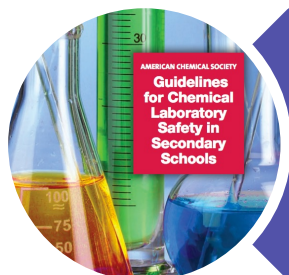
Recognizing the professional opportunities related to EHS skills

ACS Safety Programs



Division of Chemical Health and Safety (1979)

- National and Regional Meeting technical programs
- *J of Chemical Health and Safety* and DCHAS-L e-mail list
- Professional Development Workshops
- Innovative Project Grants
- Technical division partnerships, particularly CHED and CINP



Committee on Chemical Safety (1963)

- Education Subcommittee
- Communication Subcommittee
- Safe Practices Subcommittee
- Safety Advisory Panel



ACS Safety Program Office (2017)

- NSTA outreach
- CPSC support on flame-jetting education
- ACS Regional meeting workshops
- Document library maintenance
- Support for ACS outreach staff on safety issues

Key Skills Identified by a Colleague at the University of Sonora, Mexico

What 3 things you would rank as highest priority for an incoming international student to your research lab. What would you want them to make sure that they understood about laboratory safety?

1. Understanding the difference between **hazard** and **risk**
2. Identifying the different **hazards** during laboratory work
3. Recognizing how necessary **risks** can be controlled
4. The **responsibility** of the student to be part of the safety team in the laboratory



Changes in the ACS approach to chemical safety over the last 5 years

1. Management based on *rules* to development of a ***resilient safety system***
2. Focus of safety moves from *hazard identification* to ***risk assessment*** processes
3. Culture *based on compliance* to a ***one based on leadership and empowerment***
4. Safety as a *training topic* to an ***educational topic that involves research skills*** including **chemical information literacy, communication, teamwork and ethics** (see CPT guidelines for details)

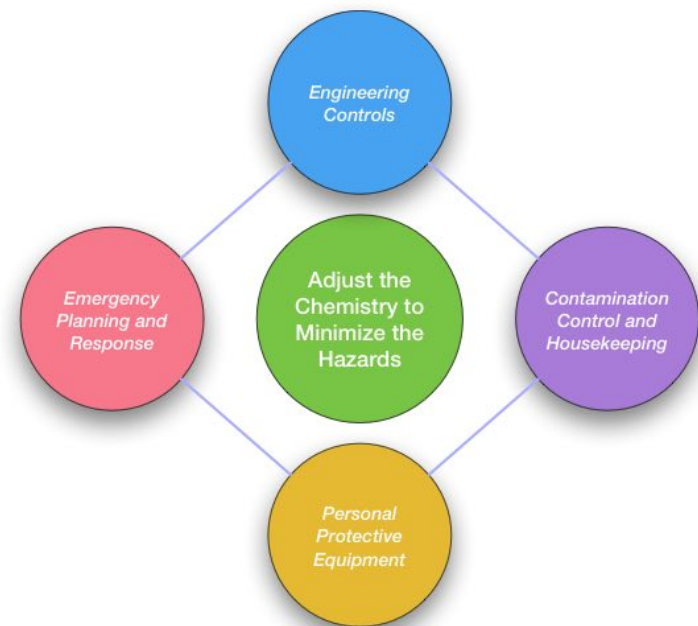


Handout:
comparison tables

The Systems Approach to Lab Safety

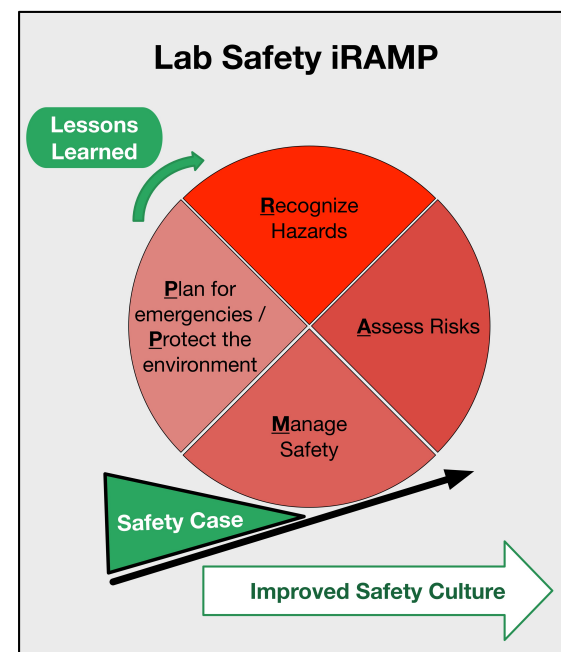
Managing chemical hazards in the lab involves coordinating 5 strategies organized into a **resilient system**:

1. Hazard Analysis and Reduction
2. Engineering Controls
3. Training and Oversight
4. Personal Protective Equipment
5. Emergency Planning and Environmental Protection



Building the System: The RAMP Chemical Safety Process

- 1. Recognize:**
What are the chemical hazards?
- 2. Assess:**
What are the most important risks?
- 3. Manage:**
How are we going to control the risks?
- 4. Prepare:**
What emergencies should we plan for?
- 5. Protect the Environment:**
How will we manage the wastes?



Selected ACS Safety Resources Since 2011

1. *Identifying and Evaluating Hazards in Research Laboratories*, (pdf) 2013
2. *CPT Guidelines for Bachelor's Degree Programs - Safety Supplement*, 2015
3. PubChem Laboratory Chemical Safety Summaries, 2015
4. *Guidelines for Chemical Laboratory Safety in Secondary Schools*, 2016
5. *Guidelines for Chemical Laboratory Safety in Academic Institutions*, 2016
6. CHED Safety Committee demonstration guidelines, 2016
7. ACS Journals Publication Policy, 2016
8. *Hazard Assessment in Research Laboratories* web site
9. *Five Key Questions for Safe Research and Demos* inChemistry, 2016
10. Bowtie symposium and article, 2016 - 2017
11. Chemistry Risk Assessment Survey, 2017
12. *Safety Guidelines for the Chemistry Professional*, 2017
13. *Safety in Academic Chemistry Labs*, 8th Edition, 2017
14. Chemical Safety webinars, 2017
15. Video evaluation rubric, in process



Links available from
<http://www.acs.org/safety>

The Developing Structure of ACS Resources

Technical

Guidelines for Chemical Laboratory Safety in Secondary Schools, 2016

Safety in Academic Chemistry Laboratories (SACL), 8th edition, 2017

Guidelines for Chemical Laboratory Safety in Academic Institutions, 2016

Hazard Assessment in Research Laboratories, 2016

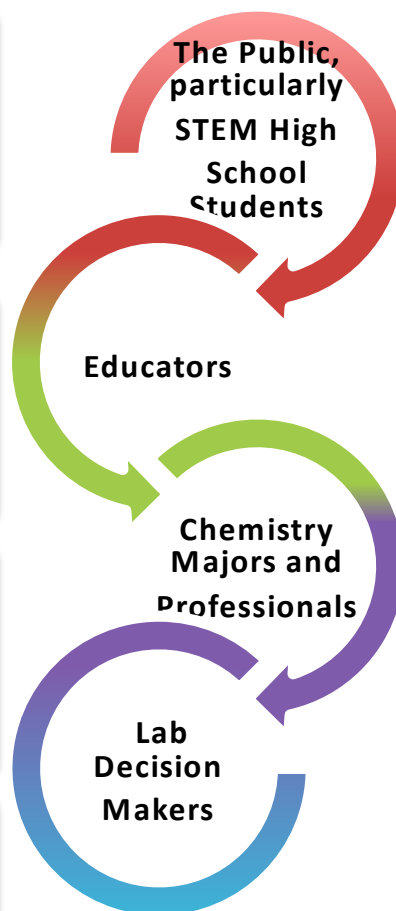
Cultural

SOCED's Guidelines and Recommendations for the Teaching of High School Chemistry, 2012

Guidelines for Bachelor's Degree Programs - Safety Supplement, Committee on Professional Training, 2015

Safe Science, National Research Council, 2014

ACS journals policy, 2016



Elephant's Toothpaste for the Public



Handout: demo assessment rubric

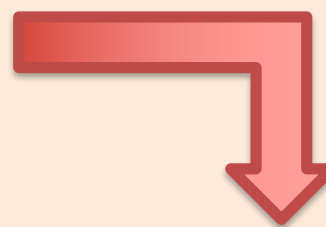
How Does the Demonstrator Do ?

In Terms of:

1. *Risk* assessment as opposed to *hazard* identification
2. *Safety system* based on resilience
3. Safety culture based on *leadership and empowerment*
4. Educational topic that involves *research skills* including chemical information literacy, communication, teamwork and ethics



Technical Chemical Safety Tools



HOW TO DO A LAB DEMO SAFELY In response to recent accidents in the classroom, here is a guide for performing experiments or demonstrations involving open flames, fire, or the use of flammables, reactives, toxic, or corrosive chemicals.

Preparing for demos or experiments:

- Determine educational goals and how the activity will meet them.
- Perform hazard and risk assessments.
- Provide a safety briefing to students.

Store bulk quantities of chemicals in a locked separate room or cupboard. Dispense only necessary quantities to sealed, suitable bottles which students access.

No hot drinks and

Use a fume hood if possible. If not, place an impact resistant screen between the demo and students. If a screen is not possible, ensure students are at least 12 feet (3.7 meters) away from the demo.

Wear appropriate personal protective equipment.

PREP **EXIT**

NOTE: Instructions in handling also need to be revised and knowledge of the safety procedures, emergency plans, the involvement in the lab, the appropriate use of personal protective equipment, and how to properly conduct a lesson risk assessment.

©2018 National Fire Protection Association Standard 45, 2018 Edition

Recognize: Managing Chemical Safety Information

- GHS is a hazard banding system – it identifies key data to assign hazard rankings for generic scenarios, thereby ignoring other data that could be relevant in some situations (e.g. auto-ignition temperature).
- The hazard rankings include concentration, but not quantity

NIOSH Pocket Guide to Chemical Hazards



SDS and Chemical Information
from Manufacturers



Right to Know Hazardous
Substance Fact Sheets

PubChem LCSS's: Safety Data based on GHS

- Safety information on 103,000 chemicals
- Includes SDS-style information as well as specific reaction information between chemicals
- All of this safety data can be downloaded and stored locally

PubChem OPEN
Laboratory Chemical Safety Summary for CID: 166

Acetone

PubChem ID: 166

Chemical Names: Acetone, 2-propanone, Propanone, Dimethyl ketone, Methyl ketone, SF-84-1

Molecular Formula: C3H6O or CC(=O)C or CC(C)=O

Molecular Weight: 58.08 g/mol

PUBCHEM > COMPOUND > ACETONE > LCSS

Contents

- 1 GHS Classification
- 2 Identifiers
- 3 Physical Properties
- 4 Toxicity Data
- 5 Exposure Limits
- 6 Health and Symptoms
- 7 First Aid
- 8 Flammability and Explosivity
- 9 Stability and Reactivity
- 10 Storage and Handling
- 11 Cleanup and Disposal
- 12 Information Sources

1 GHS Classification

Signal: **Danger**

GHS Hazard Statements

H225: Highly flammable liquid and vapor (**Danger**) Flammable liquids - Category 2

H332: Causes eye irritation (**Warning**) Serious eye damage/eye irritation - Category 2B

H333: May cause drowsiness or dizziness (**Warning**) Specific target organ toxicity, single exposure; Narcotic effects - Category 3

H360FD: Suspected of damaging fertility or the unborn child (**Warning**) Reproductive toxicity - Category 2

H412: Causes damage to organisms through prolonged or repeated exposure (**Danger**) Specific target organ toxicity, repeated exposure - Category 1

Precautionary Statement Codes

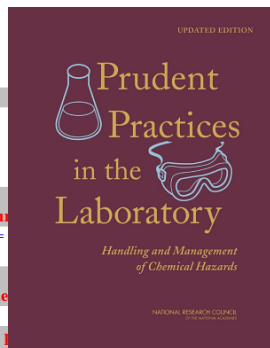
P201, P202, P210, P231, P232, P240, P241, P242, P243, P261, P273, P301, P302, P303, P304, P305, P306, P307, P308, P309, P310, P311, P312, P313, P314, P315, P321, P330, P331, P332, P333, P334, P340, P341, P342, P343, P350, P351, P353, P361, P362, P363, P370, P371, P372, P373, P374, P375, P376, P377, P378, P379, P380, P381, P382, P383, P384, P385, P386, P390, P391, P392, P393, P394, P400, P401, P402, P403, P404, P405, P406, P407, P408, P410, P411, P412, P413, P414, P415, P420, P421, P422, P423, P424, P425, P426, P427, P428, P430, P431, P432, P433, P434, P435, P436, P437, P438, P439, P440, P441, P442, P443, P444, P445, P446, P447, P448, P449, P450, P451, P452, P453, P454, P455, P456, P457, P458, P459, P460, P461, P462, P463, P464, P465, P466, P467, P468, P469, P470, P471, P472, P473, P474, P475, P476, P477, P478, P479, P480, P481, P482, P483, P484, P485, P486, P487, P488, P489, P490, P491, P492, P493, P494, P495, P496, P497, P498, P499, P500, P501, P502, P503, P504, P505, P506, P507, P508, P509, P510, P511, P512, P513, P514, P515, P516, P517, P518, P519, P520, P521, P522, P523, P524, P525, P526, P527, P528, P529, P530, P531, P532, P533, P534, P535, P536, P537, P538, P539, P540, P541, P542, P543, P544, P545, P546, P547, P548, P549, P550, P551, P552, P553, P554, P555, P556, P557, P558, P559, P560, P561, P562, P563, P564, P565, P566, P567, P568, P569, P570, P571, P572, P573, P574, P575, P576, P577, P578, P579, P580, P581, P582, P583, P584, P585, P586, P587, P588, P589, P590, P591, P592, P593, P594, P595, P596, P597, P598, P599, P600, P601, P602, P603, P604, P605, P606, P607, P608, P609, P610, P611, P612, P613, P614, P615, P616, P617, P618, P619, P620, P621, P622, P623, P624, P625, P626, P627, P628, P629, P630, P631, P632, P633, P634, P635, P636, P637, P638, P639, P640, P641, P642, P643, P644, P645, P646, P647, P648, P649, P650, P651, P652, P653, P654, P655, P656, P657, P658, P659, P660, P661, P662, P663, P664, P665, P666, P667, P668, P669, P670, P671, P672, P673, P674, P675, P676, P677, P678, P679, P680, P681, P682, P683, P684, P685, P686, P687, P688, P689, P690, P691, P692, P693, P694, P695, P696, P697, P698, P699, P700, P701, P702, P703, P704, P705, P706, P707, P708, P709, P710, P711, P712, P713, P714, P715, P716, P717, P718, P719, P720, P721, P722, P723, P724, P725, P726, P727, P728, P729, P730, P731, P732, P733, P734, P735, P736, P737, P738, P739, P740, P741, P742, P743, P744, P745, P746, P747, P748, P749, P750, P751, P752, P753, P754, P755, P756, P757, P758, P759, P760, P761, P762, P763, P764, P765, P766, P767, P768, P769, P770, P771, P772, P773, P774, P775, P776, P777, P778, P779, P780, P781, P782, P783, P784, P785, P786, P787, P788, P789, P790, P791, P792, P793, P794, P795, P796, P797, P798, P799, P800, P801, P802, P803, P804, P805, P806, P807, P808, P809, P810, P811, P812, P813, P814, P815, P816, P817, P818, P819, P820, P821, P822, P823, P824, P825, P826, P827, P828, P829, P830, P831, P832, P833, P834, P835, P836, P837, P838, P839, P840, P841, P842, P843, P844, P845, P846, P847, P848, P849, P850, P851, P852, P853, P854, P855, P856, P857, P858, P859, P860, P861, P862, P863, P864, P865, P866, P867, P868, P869, P870, P871, P872, P873, P874, P875, P876, P877, P878, P879, P880, P881, P882, P883, P884, P885, P886, P887, P888, P889, P890, P891, P892, P893, P894, P895, P896, P897, P898, P899, P900, P901, P902, P903, P904, P905, P906, P907, P908, P909, P910, P911, P912, P913, P914, P915, P916, P917, P918, P919, P920, P921, P922, P923, P924, P925, P926, P927, P928, P929, P930, P931, P932, P933, P934, P935, P936, P937, P938, P939, P940, P941, P942, P943, P944, P945, P946, P947, P948, P949, P950, P951, P952, P953, P954, P955, P956, P957, P958, P959, P960, P961, P962, P963, P964, P965, P966, P967, P968, P969, P970, P971, P972, P973, P974, P975, P976, P977, P978, P979, P980, P981, P982, P983, P984, P985, P986, P987, P988, P989, P990, P991, P992, P993, P994, P995, P996, P997, P998, P999

View GHS Classification from all 12 sources

Free NITE-CMC

Organizing chemical safety data

NAME		
SYNONYMS [synonyms]	CAS#	Formula
[top header, 3.3]	[top header, 3.2.1]	[top header]
PHYSICAL PROPERTIES [experimental properties]		
Odor: [4.2.3, 4.2.19]	Appearance: [top header, 4.2.1]	
Water Solubility: [4.2.8]	Vapor Density: [4.2.10]	
Flash Point: [4.2.7]	Vapor Pressure: [4.2.11]	
Autoignition: [4.2.13]	bp/mp: [4.2.5]/ [4.2.6]	
TOXICITY [toxicological information]		
LD₅₀ oral () [10.1.8]	EXPOSURE LIMITS	
	[safety and hazard properties]	
	TLV-TWA [9.2.18], [9.2.26]	
	(ACGIH)	
LC₅₀inhal. () [10.1.8]	STEL (ACGIH) [9.2.26]	
LD₅₀ skin () [10.1.8]	PEL (OSHA) [9.2.5]	
HEALTH AND SYMPTOMS [hazard identification] [toxicological information]		
General [9.1.2-10/12], [10.1.2]		
Skin [9.1.7,12]		
Eyes [9.1.9,12]		
Ingestion [9.1.10,12]		
Inhalation [9.1.8,12]		
FIRST AID [first aid measures]		
Skin [9.3.1.5]		
Eyes [9.3.1.6]		
Ingestion [9.3.1.7]		
Inhalation [9.3.1.4]		
FLAMMABILITY & EXPLOSIVITY		
[safety and hazard properties],[fire fighting measures]		
[9.2.9] NFPA rating (flammability) = [9.2.13]; LEL = [9.4], [9.3.2.3], [9.2.16]		
REACTIVITY & INCOMPATIBILITY		
[reactivities and incompatibilities],[properties - chemical]		
[9.8], [9.2.17]		
STORAGE & HANDLING [exposure control and personal protection]		
[storage],[accidental release - other preventative measures]		
[9.7], [9.6], [9.5.4]		
CLEANUP & DISPOSAL		
[accidental release measures],[handling - nonfire spill response]		
[9.5], [9.6.1]		
ADDITIONAL CONSIDERATIONS		



PubChem OPEN CHEMISTRY DATABASE

Search Compounds

LCSS Laboratory Chemical Safety Summary for CID 180

PUBCHEM > COMPOUND > ACETONE > LCSS

Acetone

PubChem CID: 180

Chemical Names: Acetone; 2-propanone; Propanone; Dimethyl ketone; Methyl ketone; 67-64-1

Molecular Formula: C₃H₆O or CH₃-CO-CH₃ or (CH₃)₂CO

Molecular Weight: 58.0794 g/mol

Contents

- 1 GHS Classification
- 2 Identifiers
- 3 Physical Properties
- 4 Toxicity Data
- 5 Exposure Limits
- 6 Health and Symptoms
- 7 First Aid
- 8 Flammability and Explosivity
- 9 Stability and Reactivity
- 10 Storage and Handling
- 11 Cleanup and Disposal
- 12 Information Sources

GHS Classification

Signal: **Danger**

GHS Hazard Statements

H225: Highly Flammable liquid and vapor [Danger Flammable liquids - Category 2]

H314: Causes serious eye irritation [Warning Serious eye damage/eye irritation - Category 2A]

H336: May cause drowsiness or dizziness [Warning Specific target organ toxicity, single exposure; Narcotic effects - Category 3]

Precautionary Statements

P210: Keep away from heat, hot surface, sparks, open flames and other ignition sources. - No smoking.

P233: Keep container tightly closed.

P240: Ground/bond container and receiving equipment.

PubChem LCSS and Information Literacy

PubChem LCSS Laboratory Chemical Safety Summary for CID 180

Acetone 3.3 Boiling Point

PubChem CID: 180
Chemical Names: Acetone
Molecular Formula: CC(=O)C
Molecular Weight: 58.08

56.05 deg C at 760 mm Hg
from HSDB

56 °C
from ILO-ICSC

133 °F
from NIOSH-PocketGuide, OSHA Occupational Chemical DB

133 °F (at 760 mmHg)
(NTP, 1992)
from CAMEO Chemicals

Examples

- Acetone
- Benzene
- Ethanol
- Formaldehyde
- Hydrogen Cyanide
- Imidazole
- Phenolphthalein
- Phosphoric Acid
- Theophylline
- Toluene

See a list of all compounds with LCSS

3.3 Boiling Point

56.05 deg C at 760 mm Hg
from HSDB

56 °C
from ILO-ICSC

133 °F
from NIOSH-PocketGuide, OSHA Occupational Chemical DB

Source Name: NIOSH-PocketGuide
Source ID: npgd0004
Record Name: Acetone
URL: <http://www.cdc.gov/niosh/npg/npgd0004.html>

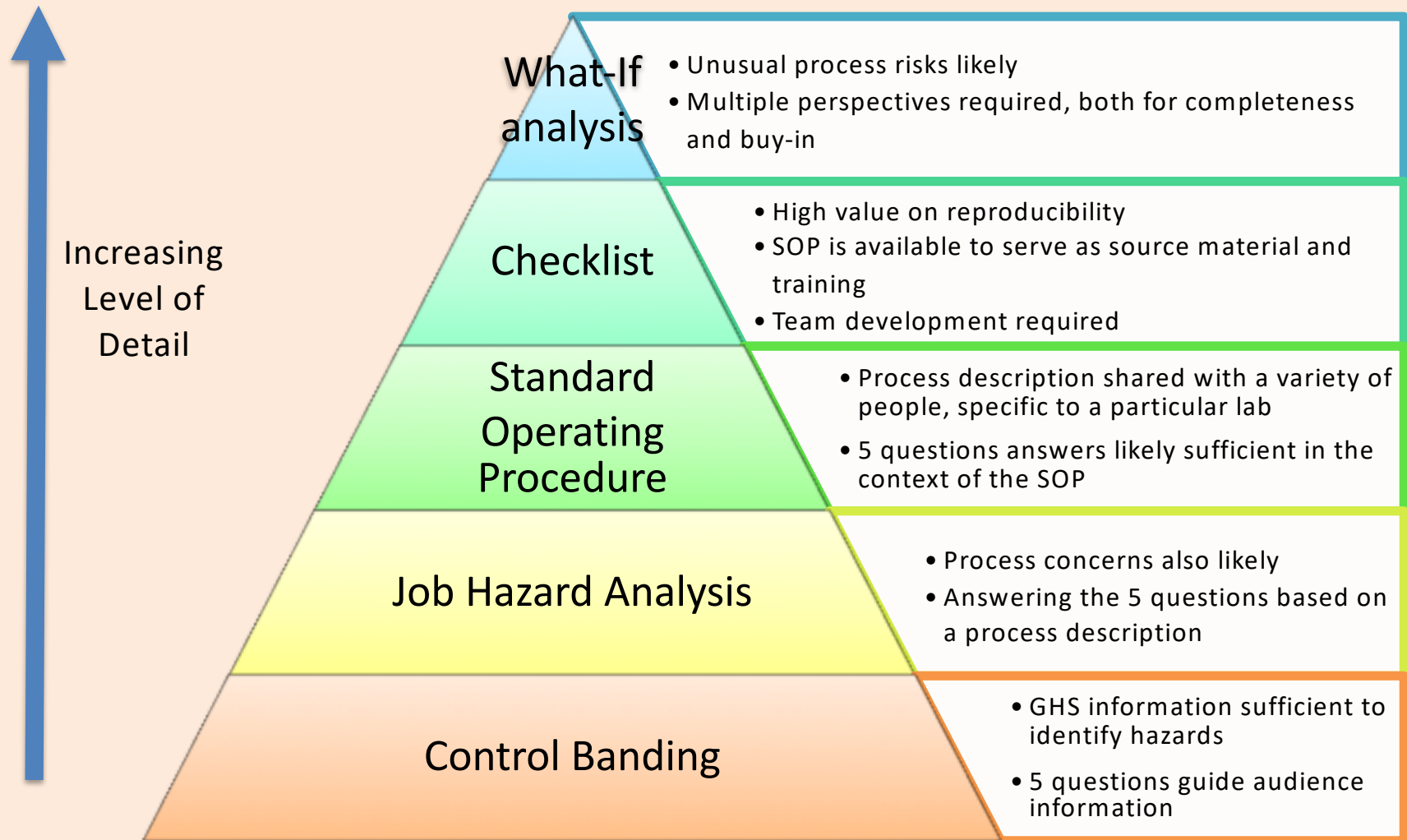
Source Name: OSHA Occupational Chemical DB
Source ID: 476
Record Name: ACETONE
URL: <http://www.osha.gov/chemicaldata/chemResult.html?RecNo=476>

133 °F (at 760 mmHg)
(NTP, 1992)
from CAMEO Chemicals

LCSS consolidates available health and safety data

- enables rapid cross examination (agreement?)
- fills in information gaps between resources
- provenance clear (with URL to data source)

Assess: Identifying and Evaluating Hazards in Research Laboratories Web Site



Manage: Building a Control-Banded Safety System

Five questions that everyone in the lab should be able to answer:

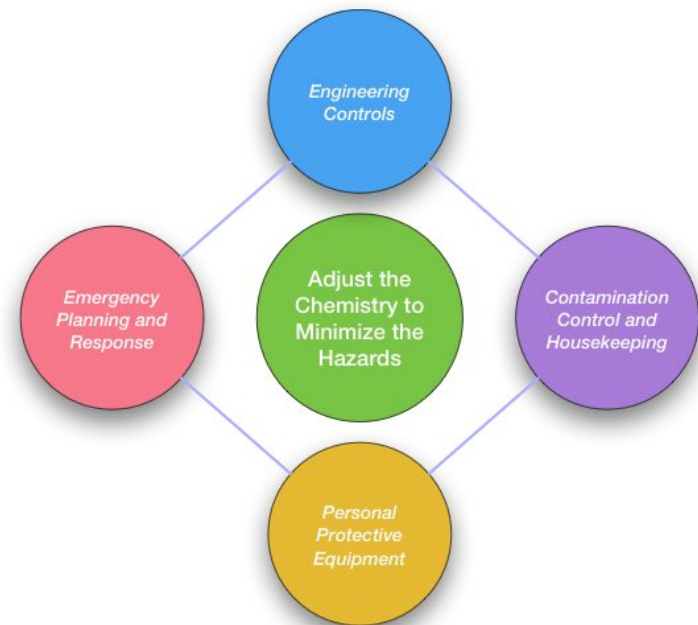
1. What are the most important chemical (GHS) and process (physical) **hazards** associated with this work?
2. What **ventilation** is required for this work and why?
3. What **Personal Protective Equipment** is required for this work and why?
4. What **emergencies** should we be ready for?
5. What **wastes** will be generated and where will they go?

		IMPACT		
		Negligible	Harmful	Serious
LIKELIHOOD	Very likely/ frequent	Moderate risk	High risk	High risk
	Somewhat likely	Low risk	Moderate risk	High risk
	Unlikely	Low risk	Low risk	Moderate risk

Lab Safety requires a System, not a Solution

Managing chemical hazards in the lab involves coordinating 5 strategies organized into a **resilient system**:






1. Hazard Analysis and Reduction
2. Engineering Controls
3. Training and Oversight
4. Personal Protective Equipment
5. Emergency Planning and Environmental Protection








Handout 5 questions article and editorial

Question 1: What are the Chemical and Process Hazards?

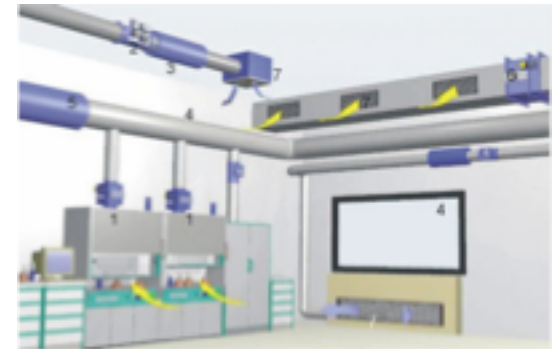
The GHS Pictograms, Signal Words and Hazard Statements identify chemical hazards associated with your work. Look especially for the “**DANGER**” signal word to identify high hazard chemicals – these are chemicals that require special planning.

Physical Hazards		
Pictogram	GHS class	Signal Words
	Explosive	<i>Danger or Warning</i>
	Oxidizer	<i>Danger or Warning</i>
	Flammable	<i>Danger or Warning</i>
	Corrosive	<i>Warning only (physical)</i>
	Compressed Gas	<i>Warning only</i>

Health Hazards		
Pictogram	GHS class	Signal Words
	Corrosive	<i>Danger only (health)</i>
	Toxic	<i>Danger only</i>
	Health Hazard	<i>Danger or Warning</i>
	Irritant	<i>Warning only</i>
	Environmental	<i>Warning only</i>

Question 2: What Ventilation Do I Need?

How much ventilation you need will depend on the fire and toxicity hazards associated with the demonstration or experiment.



“Lab ventilation” means that there is no air recirculated. The amount of room ventilation can vary:

1. **No Lab Ventilation Required** (0-3 air changes/hour)
2. **General Lab Ventilation with appropriate volatile chemical storage** (6 or more air changes/hour)
3. **Local Ventilation or Fume Hood** (>40 ACH, used for gasses)
4. **Outdoor Settings** (variable air changes, dependent on wind speed and direction)

Question 3: What PPE Do I Need?

Selecting Personal Protective Equipment (PPE) requires balancing three factors:

1. The hazards of the chemicals used
2. The scenario of concern (e.g. spill, incidental contact, contamination control)
3. The ergonomics of the PPE for the person using it: fit, dexterity, fatigue

FITS LIKE A GLOVE
Choosing the Right Glove for the Job

What part of your body is most exposed to chemicals in lab work and demos?
Your hands, of course!
Protecting them with gloves is quick and easy, if you know which type to use.

POLYETHYLENE
Advantages:
• Excellent protection from common acids and bases
Disadvantages:
• Inexpensive
• Limited tear resistance
Good protection from:
• Acids
• Detergents
• Common dilute lab reagents
Poor protection from:
• Concentrated reagents and solvents

NEOPRENE
Advantages:
• High density
• Tear resistant
Disadvantages:
• Impaired dexterity
Good protection from:
• Peroxides
• Fuels
• Alcohols
• Organic acids and bases
Poor protection from:
• Halogenated compounds
• Aromatic compounds

NITRILE
Advantages:
• Flexible
• Sturdy
• Easy to see punctures
Disadvantages:
• Limited chemical protection
Good protection from:
• Oils and greases
• Acids, caustics
• Alcohols
• Chlorinated solvents
Poor protection from:
• Strong oxidizing agents
• Aromatic solvents
• Ketones
• Acetates

BUTYL
Advantages:
• Sturdy
• Reusable
Disadvantages:
• Limited sizes
• Impaired dexterity
Good protection from:
• Peroxides
• Strong acids and bases
• Alcohols
• Aldehydes
• Ketones
• Esters
• Nitro compounds
Poor protection from:
• Hydrocarbons (aliphatic, aromatic)
• Halogenated solvents

LAMINATE FILM
Advantages:
• Protection from a wide variety of chemicals
• Can be a liner under other gloves
• Good dexterity
• Good for hazmat work
Disadvantages:
• Not puncture-resistant
Good protection from:
• Alcohols
• Hydrocarbons (aliphatic, aromatic)
• Chlorines
• Ketones
• Esters
Poor protection from:
• Check manufacturer information

ACS & YOU

Individual brands vary. Always check glove compatibility against the manufacturer's recommendations.
Special thanks to the ACS Committee on Chemical Safety.
References:
www.casha.gov/Publications/casha3151.html
www.eric.bekeley.edu/workplace-safety/glove-selection-guide

Question 4: What Emergencies Should I Plan For?

Develop Emergency Plans for:

- Fires
- Medical emergencies
- Hazmat spills
- To account for cultural challenges



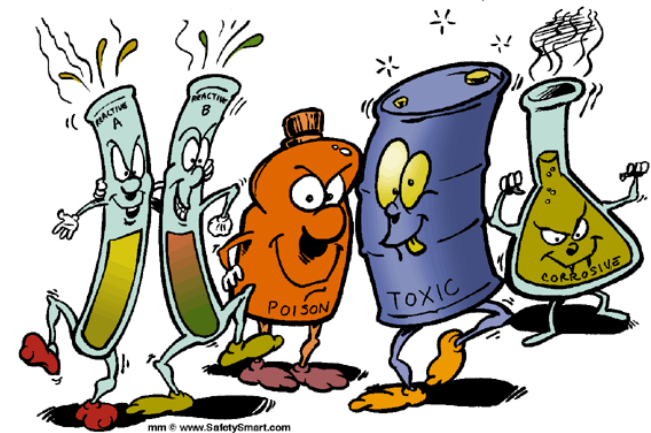
All plans should be coordinated with local response agencies.

Question 5: What Will I Do With Wastes?

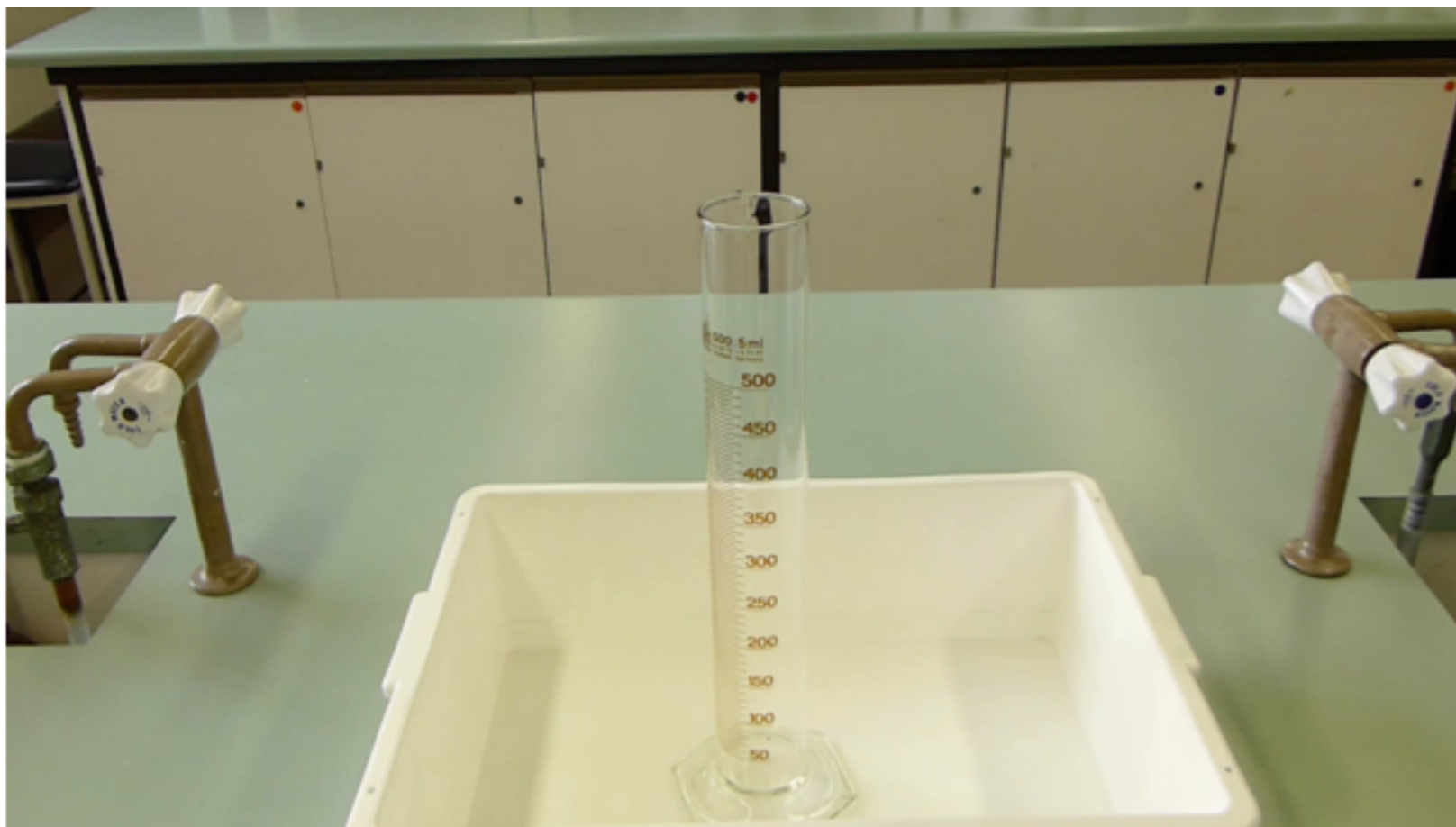
It is important to check with the host institution before the work is done to know what waste streams they are prepared to accept

Consider These Wastes:

- Chemicals
- Biological materials
- Contaminated lab materials
- Broken glassware
- General trash & recycling



Elephant's Toothpaste



Handout: J Chem Ed article and JHA

How Does the Chemist Do ?

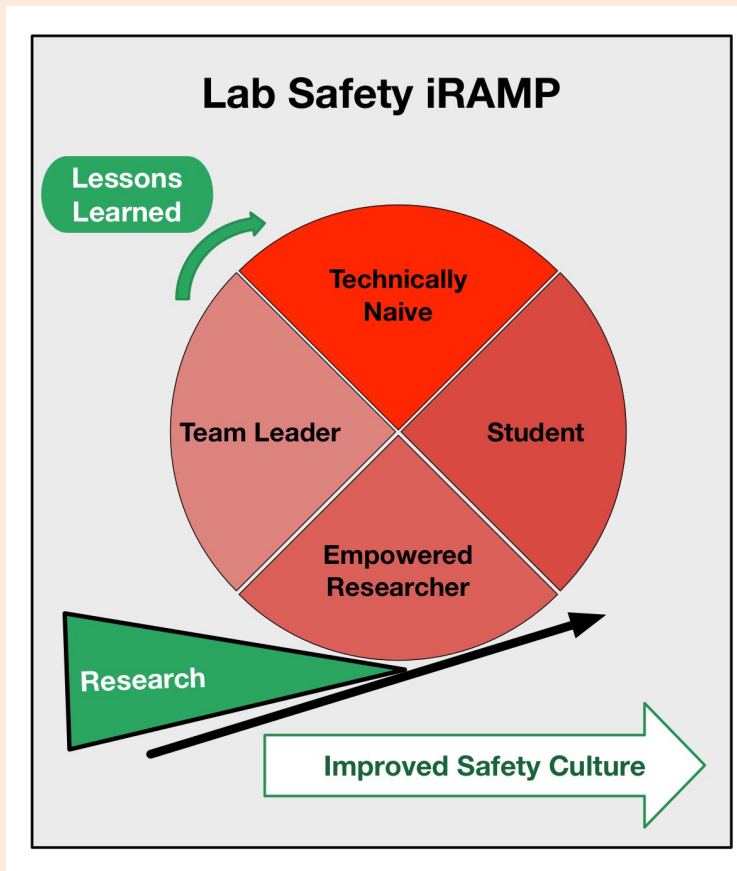
1. Risk assessment as opposed to hazard identification
2. Safety system based on resilience
3. Safety culture based on leadership and empowerment
4. Educational topic that involves research skills including chemical information literacy, communication, teamwork and ethics



RAMP Tools

Stage	Technical	Cultural	Key Stakeholder
Recognize	PubChem LCSS	Prudent Practices	Lab workers
Assess	Hazard Assessment in Research Laboratories	Risk assessment practices survey	Laboratory supervisor
Manage	5 questions; Guidelines documents and SACL 8	<i>Safe Science</i> and APLU documents	Lab workers
Plan / Protect	Regulatory compliance guidance	Public outreach and education around incidents	Institutional programs
Lessons Learned / Information Sharing	Bowtie development	Lessons Learned	Laboratory supervisor

Developing Lessons Learned



How People Learn: Brain, Mind, Experience, and School: Expanded Edition

Committee on Developments in the Science of Learning with additional material from the Committee on Learning Research and Educational Practice, National Research Council

ISBN: 0-309-50145-8, 285 pages, 7 x 10, (2000)

This PDF is available from the National Academies Press at: <http://www.nap.edu/catalog/9653.html>

Experts organize knowledge and approach problems differently from novices

Our goal: to help students build coherent framework of useful knowledge – that can be transferred to new situations

Goals of a Lessons Learned Program

- Technical
 - To avoid the **same incident**
 - To help **lateral thinking** to develop “what if” scenarios
 - To improve **emergency planning** for response to laboratory events
 - To identify **successful** health and safety **protection measures**
- Cultural
 - To help lab workers **prioritize** their safety concerns.
 - To enhance situational **awareness** of lab workers
 - To provide stories that support **safety training** efforts



Novak, J. D. A Theory of Education, Cornell University: Ithaca, 1977.

The Challenge of the "Root Cause"

From Scott Geller's **Are You a Safety Bully?**

- A common myth in safety holds that injuries are caused by one critical factor, the root cause; Ask enough questions and you'll arrive at the critical factor behind an injury.
- Is it really possible that a single root cause is responsible for a mishap...?
- Conducting an investigation to find a singular root cause could be considered bullying. This approach can put employees on the defensive, even preventing the disclosure of hazards or barriers to safe work practices.



The Difference between Incident Investigations and Lessons Learned

Incident Investigations:

(specific to an event)

- Focus on establishing "facts" to assess losses and plan recovery
- Should occur as soon as possible, before individuals develop their own interpretation of the event
- Are specific to the event at hand
- Are administratively and regulatorily mandated

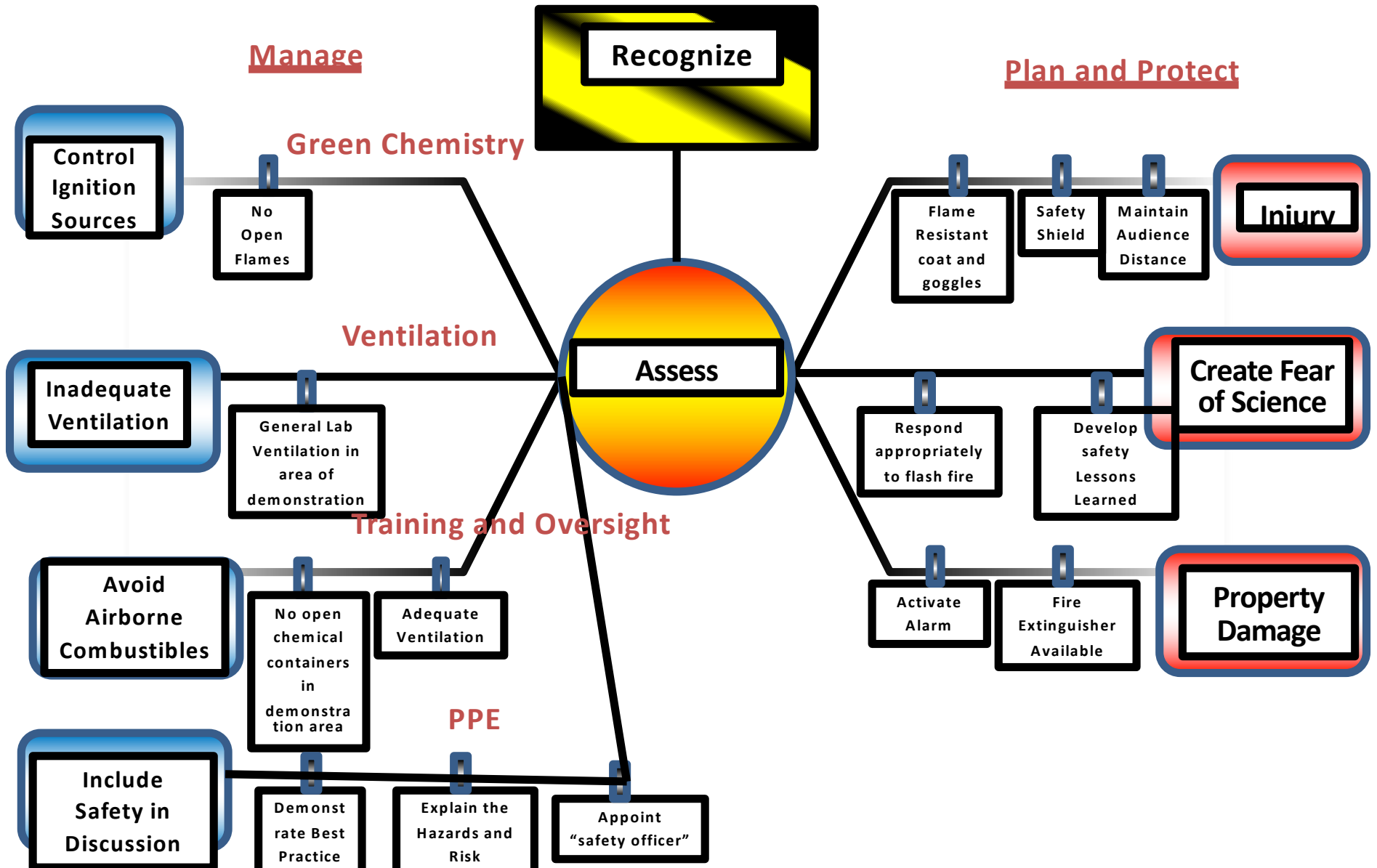


Lessons Learned:

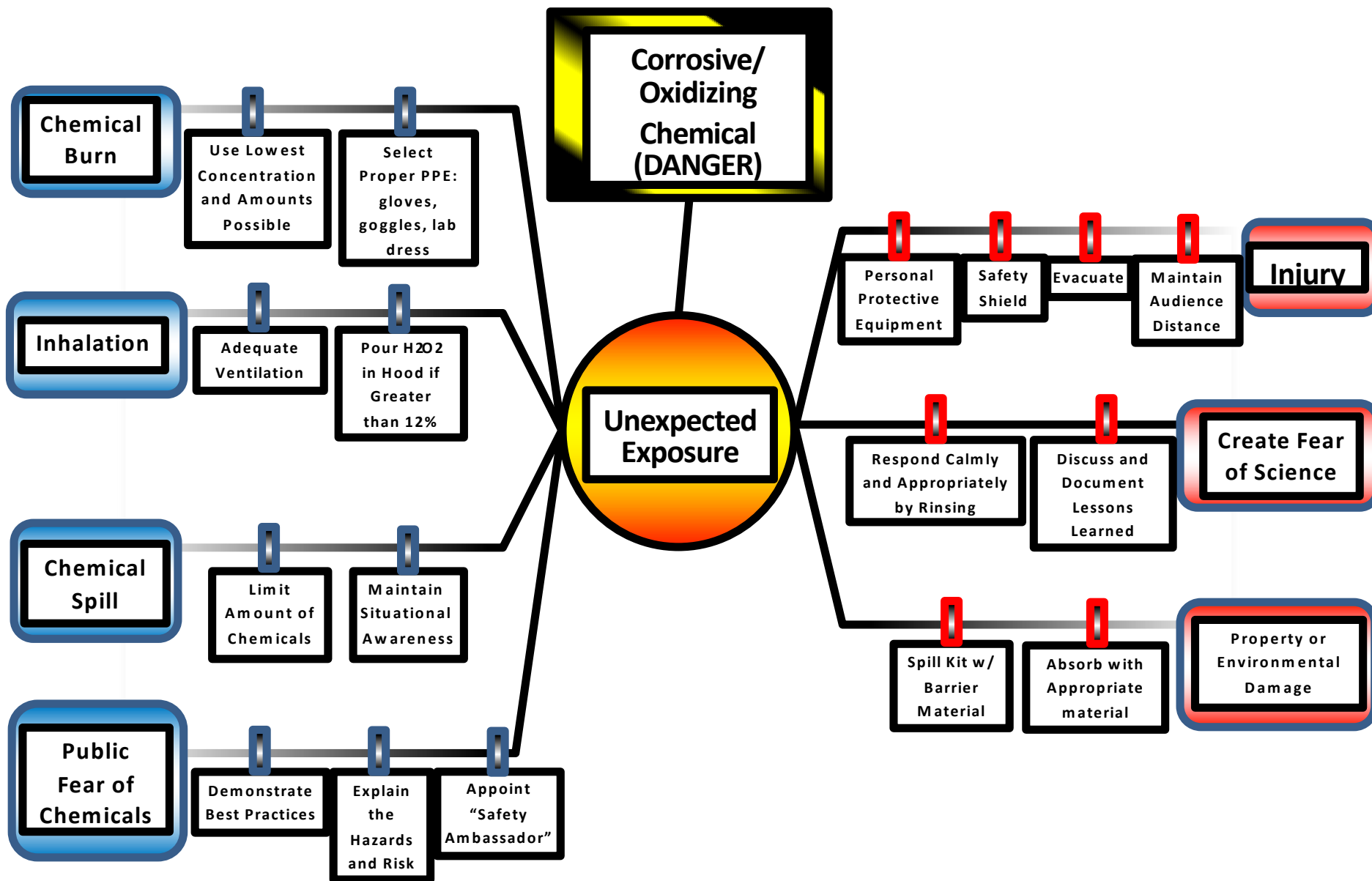
(useful for other people)

- Occur after the critical incident response phase
- Requires an emotional distance from the event
- Include emotional elements
- Priority of Lessons will vary between readers
- Can be used in other settings or for other events
- Voluntary and anonymous(?)

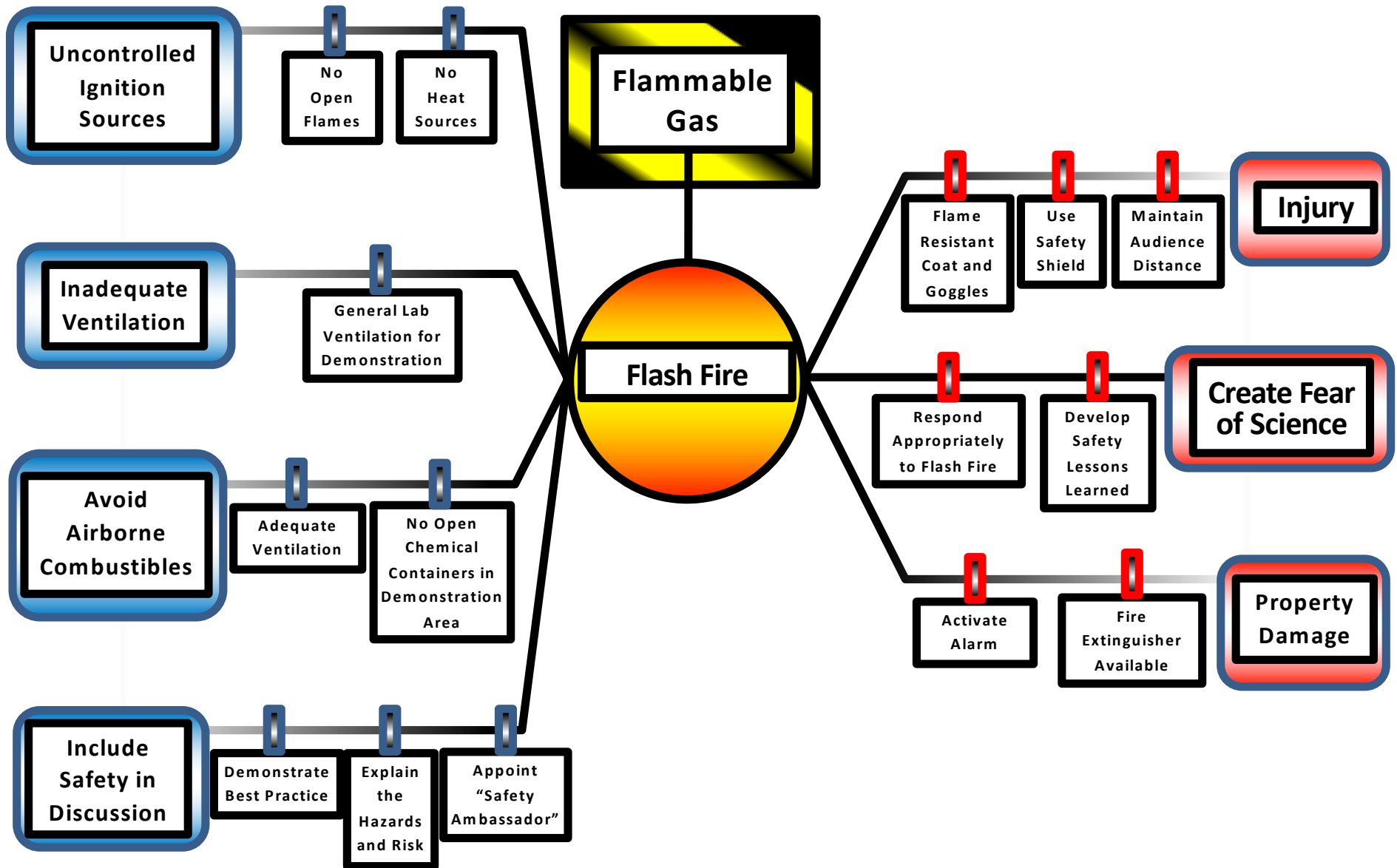
Generic Lab Bowtie



Risk Scenario One: Exposure to Hydrogen Peroxide



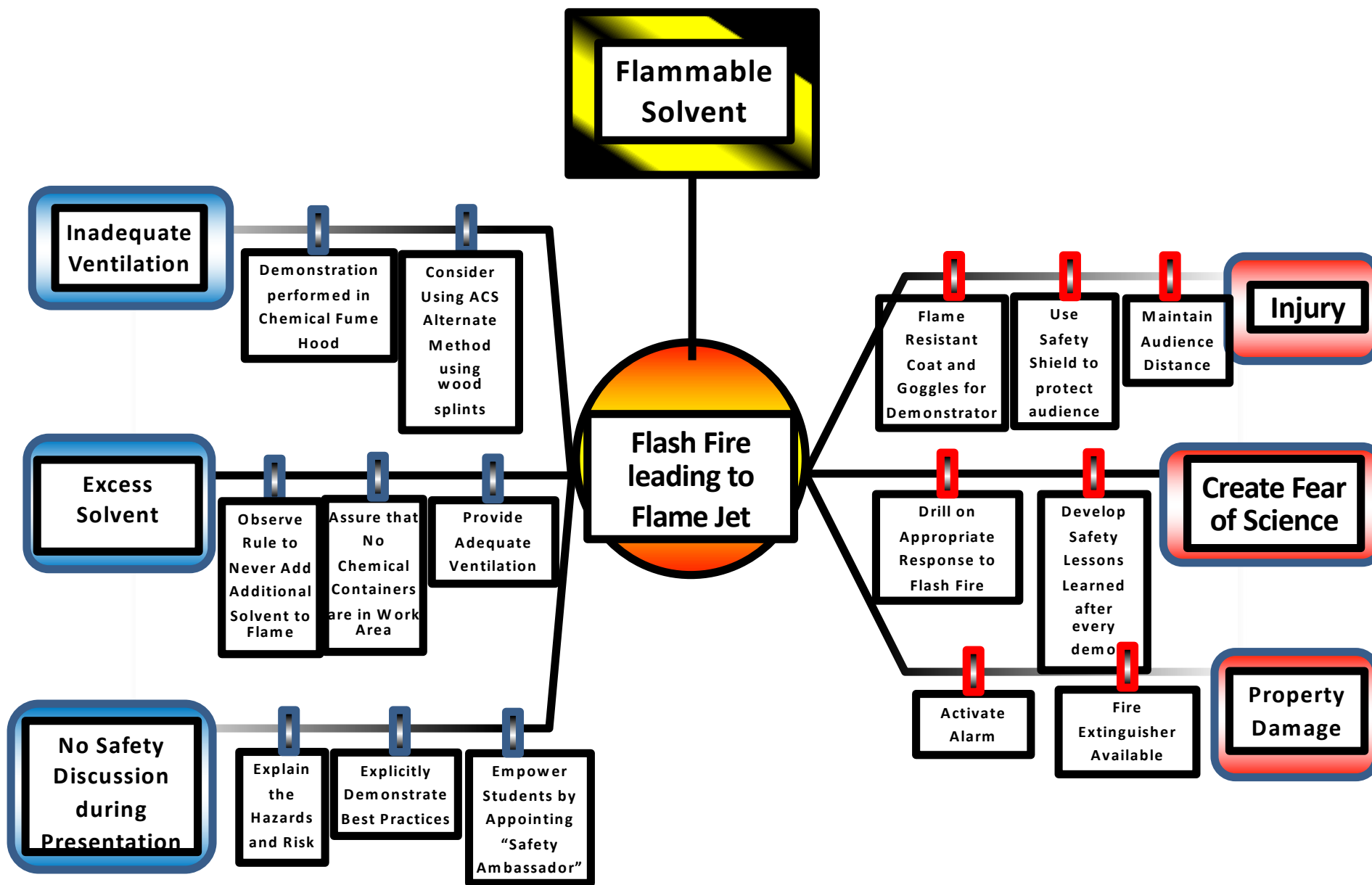
Risk Scenario: Ignition of Oxygen Gas Bubbles



After the Rainbow



Bowtie Diagram of Rainbow Demos



Another Approach to Lessons Learned

The Importance of
Situational Awareness &
Face Protection

Lessons Learned from this Video

1. What Joseph Learned
 - PPE is situation-specific
 - One size doesn't fit all scenarios
2. What I learned
 - The UCLA fire had social ripple effects as well as technical ones
 - What “**Upper Management Support**” Means in Academia
 - Look for Buried Lessons in a Story: emergency response



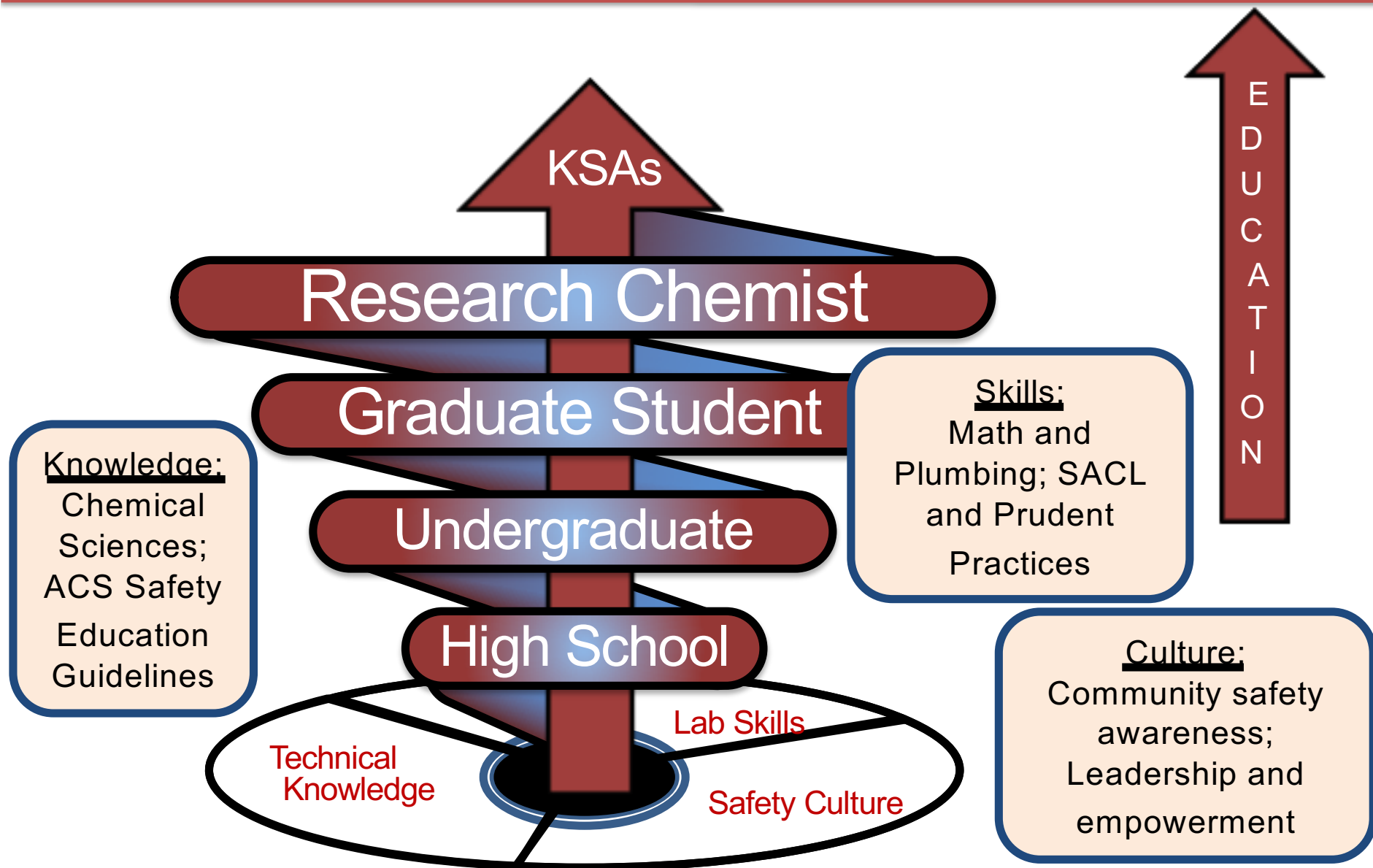
Another Form of Education



Skin exposure to 30% H₂O₂



Emerging Educational Expectations for Lab Chemical Safety



Chemical Safety References

Audience	Technical Resources	Cultural Resources
High school and undergraduate teaching labs	<u>Guidelines for Chemical Laboratory Safety in Secondary Schools</u> , ACS 2016	<u>Guidelines and Recommendations for the Teaching of High School Chemistry</u> , ACS 2012
Mentored research labs (REU, CURE, similar programs)	<u>Safety in Academic Chemistry Laboratories</u> (SACL), 8 th edition ACS 2017	<u>Creating Safety Cultures in Academic Institutions</u> ACS, 2013
Supervised research (graduate school)	<u>Guidelines for Chemical Lab Safety in Academia</u> , ACS 2016	<u>A Guide to Implementing a Safety Culture in our Universities</u> APLU, 2016
Research leadership	<u>Prudent Practices in the Laboratory</u> National Academies Press 2011 <u>Hazard Assessment in Research Laboratories</u>	<u>Safe Science</u> National Academies Press, 2014 Safety Guidelines for the Chemistry Professional ACS DCHAS / CCS, 2017

My Lessons Learned

- Connect to the Mission (**Teaching**, Research, Service) rather than the regulations
- Science in the 21st Century is changing so fast that it requires a resilient approach to safety
- Connections with other chemistry support roles are valuable – information professionals, educators
 - Leverage their skills
 - Recognize their restraints



Feedback about ACS Priorities?

Safety Program Ideas under consideration for ACS development

1. Publish the demonstration / video safety rubric
2. Organize safety information-literate classroom exercises for chemical process risk assessment
3. Flesh out the bowtie model for the lab chemical safety use case
4. Foster a Lessons Learned story collection
5. Develop chemical safety learning objectives for various laboratory skill levels
6. RAMP marketing materials for various audiences
7. Build a “crosswalk map” between chemical safety and other lab safety skills e.g. biosafety, rad safety, physical safety. (A *crosswalk* maps the elements in one schema to the equivalent elements in another schema.)
8. Others?

