



Safety in Research Articles

A Tale of Two Times

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On safety warnings

- “Review and analysis of safety policies of chemical journals” Grabowski, LE, Goode, SR, *Journal of Chemical Health & Safety*, May/June **2016**, 30-35
- “...injuries could be avoided by adding a few cautionary words to the procedure.”
- “Safety as a Core Value” Langerman, N, *Journal of Chemical Health & Safety*, May/June **2016**, 48-49
- “Safety and Ethics” Langerman, N, *Journal of Chemical Health & Safety*, May/June **2015**, 44-45



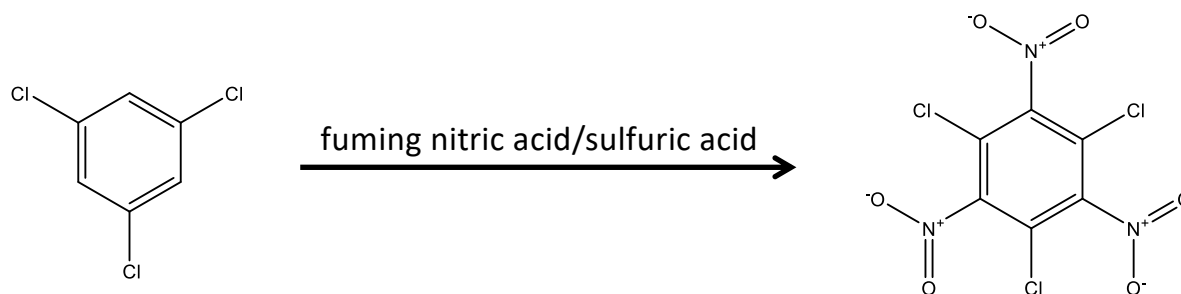
The Quest...

- Is benzene always planar?
 - 1987/88 tale (*predates Lab standard & Cal/OSHA enforcement at UC*)
 - 2018 tale...

Creating localized pi-bonds in benzene

- Electrophilic aromatic substitution
 - trichlorobenzene, fuming nitric acid, sulfuric acid
- Nucleophilic aromatic substitution
 - ammonia
- Alkylation
 - Dimethylamine, ethanol, chloroform, hexane
 - Recrystallization, hexanes, ethanol
 - Diethylamine, ethanol, chloroform, hexane

On the road to non-planar benzene



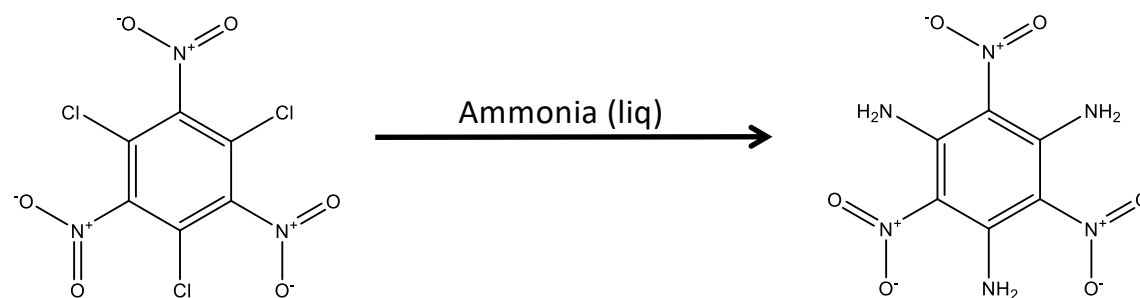
Reference:

Hill, M.E.; Taylor, F. J. Org. Chem. 1960, 25, 1037

Hazards?

Non-planar benzene next step:

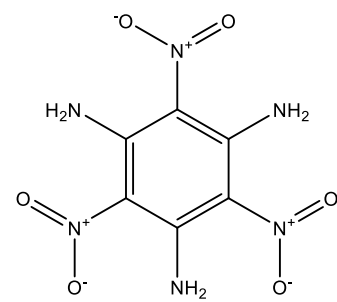
Nucleophilic aromatic substitution



Reference:

Chance, JM, et al. *J. Am. Chem. Soc.* **1989**, *111*, 5944-5946

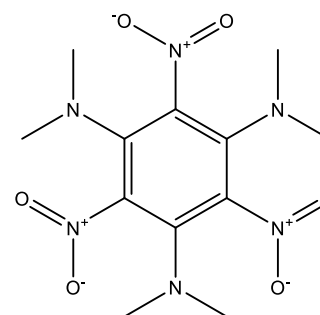
Non-planar benzene next step: Alkylation



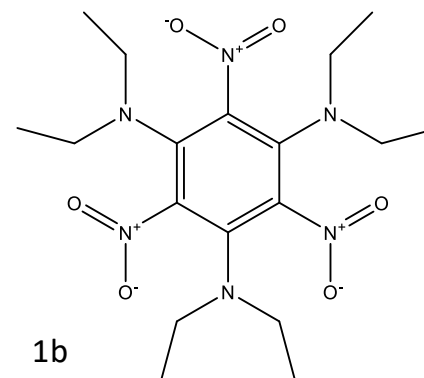
1c

dimethylamine/ethanol

diethylamine/ethanol



1a



1b

Published Hazards?

- Chance, JM, et al. *J. Am. Chem. Soc.* **1989**, *111*, 5944-5946

“Tris(dialkylamino)trinitrobenzenes are, in general, explosive! Extreme care should be taken when handling these materials. Do not prepare or store large quantities of these compounds. It is unadvisable to store these compounds in ground glass stoppered vials. Even though we have had no accidental detonations of 1a or 1b, we have been able to detonate a few milligrams of 1a by striking it with a ball peen hammer on a hard surface.”



What about today?

- GHS Hazard codes and statements part of information about each chemical
 - Included in SDS
 - Inform precautions
 - Communicate in publications

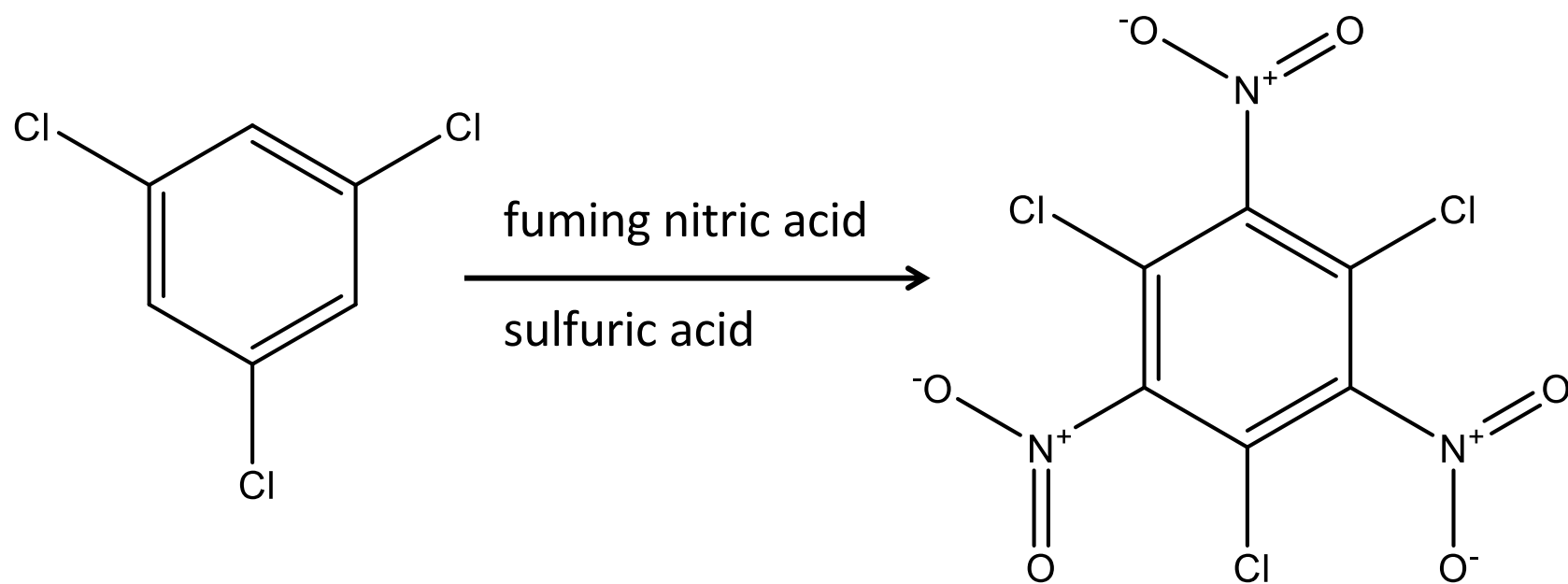
- Chemicals (developed by Risk & Safety Solutions, University of California)

Booth 525

RiskandSafetySolutions.com

- For novel compounds, no hazard codes exist. The explosive nature of the trinitro-triaminobenzene compounds cannot be extracted from reactant Hcodes

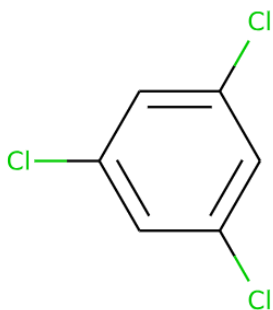
Electrophilic aromatic substitution



Electrophilic aromatic substitution hazards: trichlorobenzene

1,3,5-Trichlorobenzene

Toxic : Inhalation



CAS #

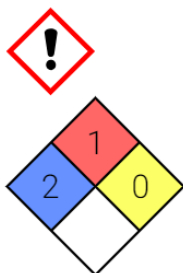
108-70-3

Molecular Formula

$C_6H_3Cl_3$

Physical State

solid



Chemical Information

Molecular Weight

181.45 g/mol

Boiling Point

208 to 208 deg. C

Flash Point

107 deg. C

Melting Point

56 to 60 deg. C

Hazard Statements



H302 - Harmful if swallowed (Category 4)

H312 - Harmful in contact with skin (Category 4)

H315 - Causes skin irritation (Category 2)

H319 - Causes serious eye irritation (Category 2A)

H332 - Harmful if inhaled (Category 4)

H335 - May cause respiratory irritation (Category 3)

H402 - Harmful to aquatic life (Category 3)

H412 - Harmful to aquatic life with long lasting effects (Category 3)

Electrophilic aromatic substitution hazards: fuming nitric acid

Nitric acid reagent grade, fuming, >90%

CFATS : Release

CFATS : Theft

Corrosive

Extremely Hazardous Substance

Oxidizers : Class 3

CAS #

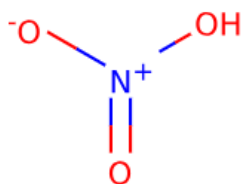
7732-18-5, 7697-37-2, 10544-72-6

Molecular Formula

HNO₃

Physical State

liquid



Chemical Information

Molecular Weight

63.01

Boiling Point

120.5 to 120.5

Flash Point

N/A

Melting Point

N/A

Hazard Statements



H272 - May intensify fire; oxidizer. (Category 2)



H290 - May be corrosive to metals (Category 1)

H318 - Causes serious eye damage (Category 1)

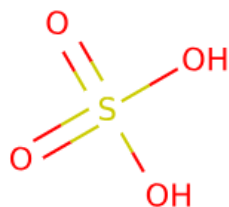
H314 - Causes severe skin burns and eye damage. (Sub-category 1A)

H333 - May be harmful if inhaled (Category 5)

Electrophilic aromatic substitution hazards: sulfuric acid

Sulfuric acid

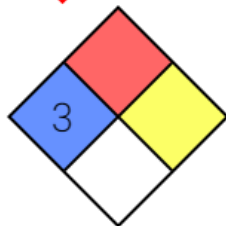
Corrosive
Extremely Hazardous Substance



CAS #
7664-93-9

Molecular Formula
N/A

Physical State
liquid



Chemical Information

Molecular Weight
98.08 g/mol

Boiling Point
290 to 290

Flash Point
N/A

Melting Point
N/A

Hazard Statements



H314 - Causes severe skin burns and eye damage. (Sub-category 1A)

H290 - May be corrosive to metals (Category 1)

H318 - Causes serious eye damage (Category 1)

H303 - May be harmful if swallowed (Category 5)



Signal words

- **Danger** - Indicates death or serious injury will result if proper precautions are not taken.
- **Warning** - Indicates death, serious injury or property damage can result if proper precautions are not taken.
- **Caution** - Indicates some injury or property damage may result if proper precautions are not taken



Electrophilic aromatic substitution reactant hazards

By Category:

- H318 - Causes serious eye damage (Category 1)
- H314 - Causes severe skin burns and eye damage. (Sub-category 1A)
- H290 - May be corrosive to metals (Category 1)
- H315 - Causes skin irritation (Category 2)
- H272 - May intensify fire; oxidizer. (Category 2)
- H319 - Causes serious eye irritation (Category 2A)
- H335 - May cause respiratory irritation (Category 3)
- H402 - Harmful to aquatic life (Category 3)
- H412 - Harmful to aquatic life with long lasting effects (Category 3)
- H332 - Harmful if inhaled (Category 4)
- H312 - Harmful in contact with skin (Category 4)
- H302 - Harmful if swallowed (Category 4)
- H303 - May be harmful if swallowed (Category 5)
- H333 - May be harmful if inhaled (Category 5)



Thesis

- Category 1 → “Danger”
- Category 2 → “Warning”
- Category 3 → “Caution”

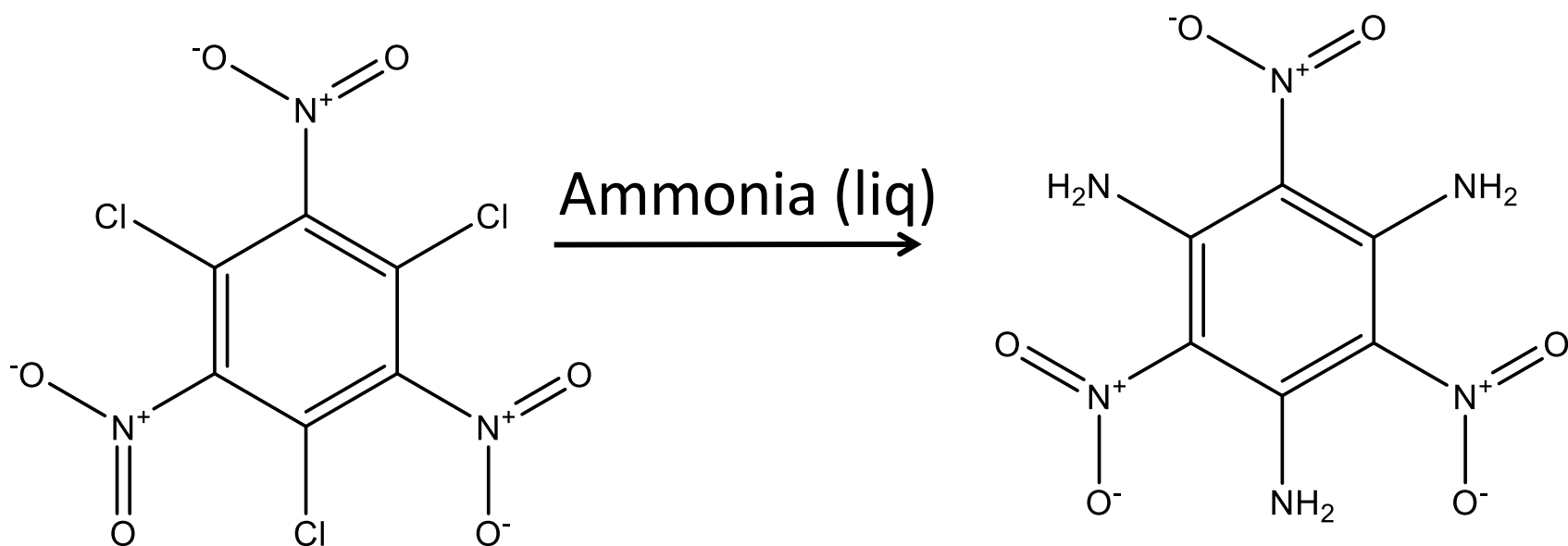


Electrophilic aromatic substitution

- **Danger:** Reactants cause serious eye and skin damage and are corrosive to metals
- **Warning:** Reactants include oxidizers which intensify fire
- **Caution:** Reactants are harmful to aquatic life

Non-planar benzene next step:

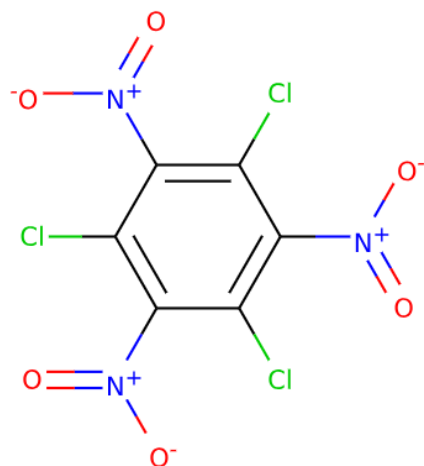
Nucleophilic aromatic substitution



Nucleophilic aromatic substitution hazards: trichloro-trinitrobenzene

1,3,5-TRICHLORO-2,4,6-TRINITRO-BENZENE

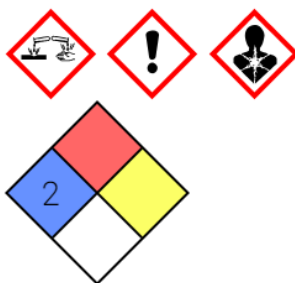
Toxic : Oral



CAS #
2631-68-7

Molecular Formula
 $C_6Cl_3N_3O_6$

Physical State
solid



Chemical Information

Molecular Weight
316.442

Boiling Point
N/A

Flash Point
N/A

Melting Point
N/A

Hazard Statements



H318 - Causes serious eye damage (Category 1)



H317 - May cause an allergic skin reaction. (Category 1)

H302 - Harmful if swallowed (Category 4)



H334 - May cause allergy or asthma symptoms or breathing difficulties if inhaled. (Category 1)

H316 - Causes mild skin irritation (Category 3)

Nucleophilic aromatic substitution hazards: ammonia

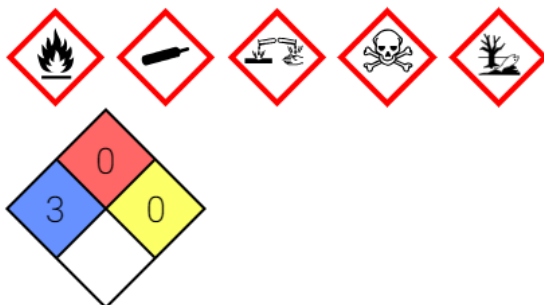
Ammonia

Corrosive
 Extremely Hazardous Substance

CAS #
 7664-41-7

Molecular Formula
 H_3N

Physical State
 gas (gaseous)



Chemical Information

Molecular Weight
 17.03 g/mol

Boiling Point
 -33 to -33 deg. C

Flash Point
 132 deg. C

Melting Point
 -78 to -78 deg. C

Hazard Statements



H221 - Flammable gas (Category 2)



H280 - Contains gas under pressure; may explode if heated (Compressed Gas)



H314 - Causes severe skin burns and eye damage (Category 1B)
 H318 - Causes serious eye damage (Category 1)



H331 - Toxic if inhaled (Category 3)



H400 - Very toxic to aquatic life (Category 1)
 H410 - Very toxic to aquatic life with long lasting effects (Category 1)



Nucleophilic aromatic substitution reactant hazards

H318 - Causes serious eye damage (Category 1)

H314 - Causes severe skin burns and eye damage (Category 1B)

H317 - May cause an allergic skin reaction. (Category 1)

H400 - Very toxic to aquatic life (Category 1)

H410 - Very toxic to aquatic life with long lasting effects (Category 1)

H334 - May cause allergy or asthma symptoms or breathing difficulties if inhaled. (Category 1)

H221 - Flammable gas (Category 2)

H331 - Toxic if inhaled (Category 3)

H316 - Causes mild skin irritation (Category 3)

H302 - Harmful if swallowed (Category 4)

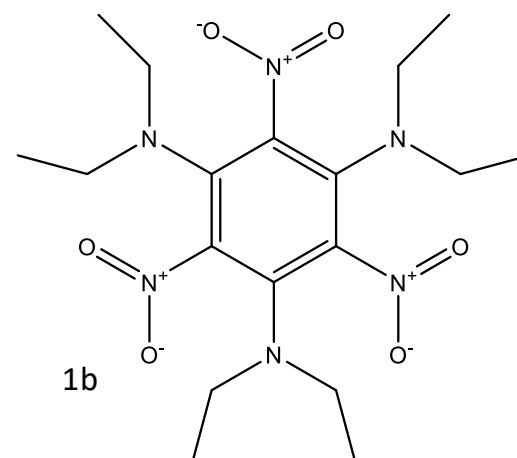
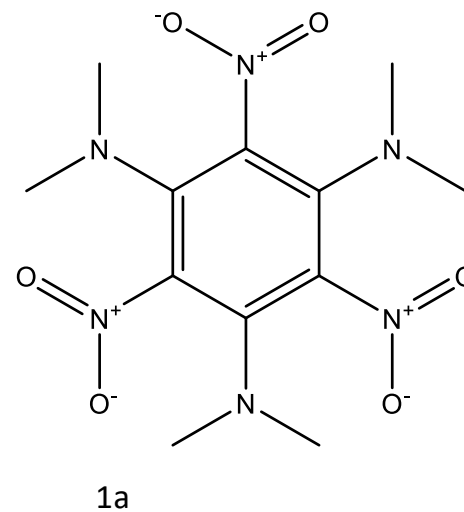
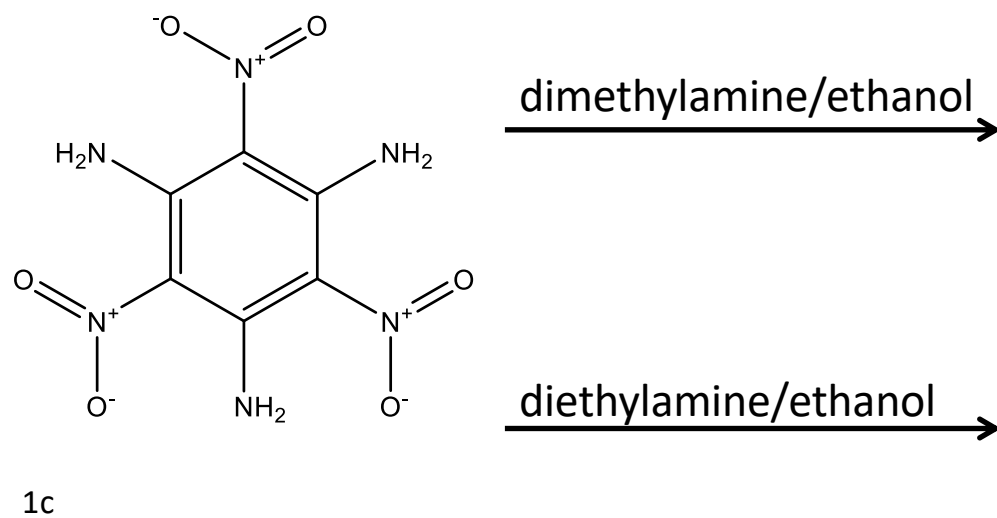
H280 - Contains gas under pressure; may explode if heated (Compressed Gas)



Nucleophilic aromatic substitution

- **Danger:** Reactants cause serious eye damage & skin burns. , may cause an allergic skin reaction, Very toxic to aquatic life with long lasting effects. May cause allergy or asthma symptoms or breathing difficulties if inhaled.
- **Warning:** Reactants include flammable gas
- **Caution:** Reactants are toxic if inhaled and cause mild skin irritation

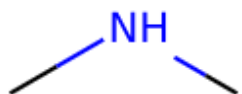
Alkylation



Alkylation hazards: dimethylamine

Dimethylamine

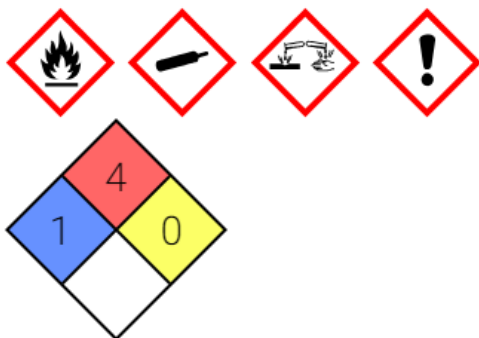
CFATS : Release
Flammable Gas
Toxic : Inhalation



CAS #
124-40-3

Molecular Formula
 C_2H_7N

Physical State
gas (gaseous)



Chemical Information

Molecular Weight
45.08 g/mol

Boiling Point
7 to 7 deg. C

Flash Point
-6.69 deg. C

Melting Point
-93 to -93 deg. C

Hazard Statements



H220 - Extremely flammable gas (Category 1)



H280 - Contains gas under pressure; may explode if heated (Liquefied Gas)



H318 - Causes serious eye damage (Category 1)



H315 - Causes skin irritation (Category 2)

H332 - Harmful if inhaled (Category 4)

H335 - May cause respiratory irritation (Category 3)

H402 - Harmful to aquatic life (Category 3)

H412 - Harmful to aquatic life with long lasting effects (Category 3)

Alkylation hazards: diethylamine

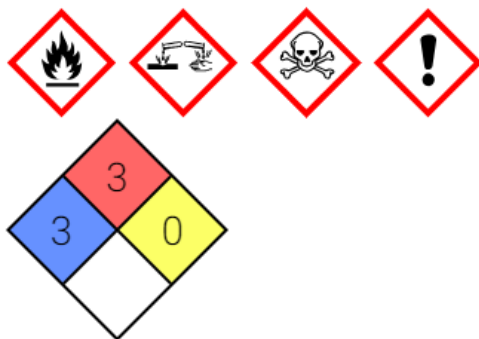
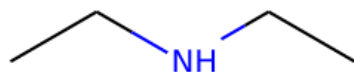
Diethylamine

Corrosive
Flammable Liquid : IB
Toxic : Inhalation
Toxic : Dermal

CAS #
109-89-7

Molecular Formula
 $C_4H_{11}N$

Physical State
liquid



Chemical Information

Molecular Weight
73.14 g/mol

Boiling Point
55 to 55 deg. C

Flash Point
-23 deg. C

Melting Point
-50 to -50 deg. C

Hazard Statements



H225 - Highly Flammable liquid and vapour (Category 2)



H314 - Causes severe skin burns and eye damage (Category 1A)
H318 - Causes serious eye damage (Category 1)



H311 - Toxic in contact with skin (Category 3)



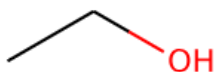
H302 - Harmful if swallowed (Category 4)
H332 - Harmful if inhaled (Category 4)
H335 - May cause respiratory irritation (Category 3)

H401 - Toxic to aquatic life (Category 2)

Alkylation hazards: ethanol

Ethanol, ≥ 190 proof

Flammable Liquid : IB



CAS #
64-17-5

Molecular Formula
 C_2H_6O

Physical State
liquid



Chemical Information

Molecular Weight
46.07 g/mol

Boiling Point
78 to 78 deg. C

Flash Point
14 deg. C

Melting Point
-114 to -114 deg. C

Hazard Statements



H225 - Highly Flammable liquid and vapour (Category 2)



H319 - Causes serious eye irritation (Category 2A)

H336 - May cause drowsiness or dizziness (Category 3)

H401 - Toxic to aquatic life (Category 2)



Alkylation hazards

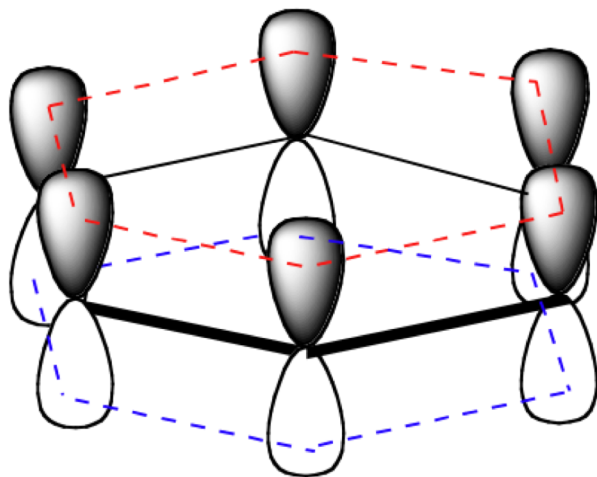
- H220 - Extremely flammable gas (Category 1)
- H314 - Causes severe skin burns and eye damage (Category 1A)
- H318 - Causes serious eye damage (Category 1)
- H225 - Highly Flammable liquid and vapor (Category 2)
- H315 - Causes skin irritation (Category 2)
- H319 - Causes serious eye irritation (Category 2A)
- H401 - Toxic to aquatic life (Category 2)
- H311 - Toxic in contact with skin (Category 3)
- H335 - May cause respiratory irritation (Category 3)
- H336 - May cause drowsiness or dizziness (Category 3)
- H402 - Harmful to aquatic life (Category 3)
- H412 - Harmful to aquatic life with long lasting effects (Category 3)
- H302 - Harmful if swallowed (Category 4)
- H332 - Harmful if inhaled (Category 4)
- H280 - Contains gas under pressure; may explode if heated (Liquefied Gas)



Alkylation Hazards

- **Danger:** Extremely flammable gas, causes severe skin burns and eye damage
- **Warning:** Highly Flammable liquid and vapor, causes skin and eye irritation, toxic in contact with skin and to aquatic life
- **Caution:** Toxic in contact with skin, may cause respiratory irritation and drowsiness or dizziness, harmful to aquatic life with long lasting effects

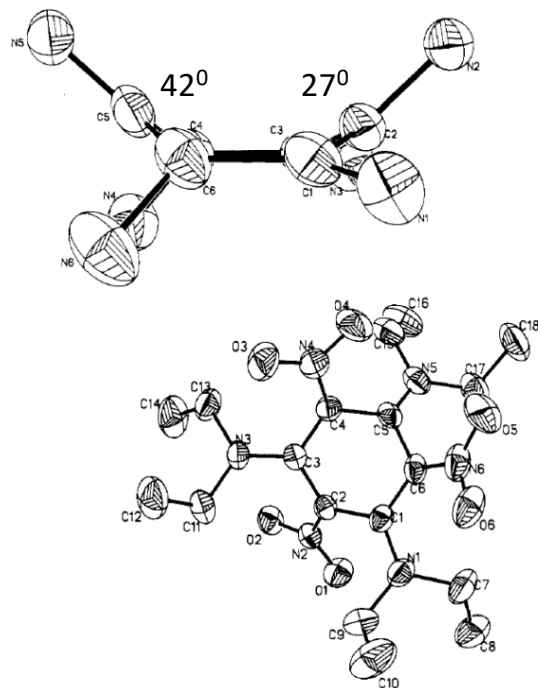
So, can you make benzene non-planar?



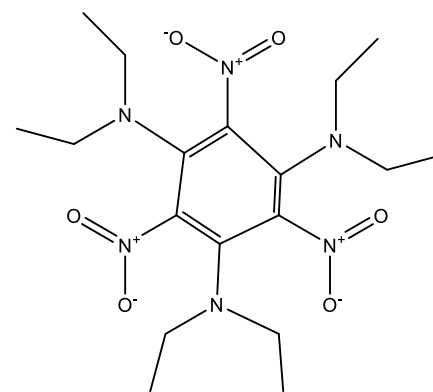
*π electrons delocalized
around the ring, above
and below the plane*

Yes

- Boat confirmation ground state



Bond	Expected	Found
C(ar)-N(ar)	1.37Å	1.34Å & 1.30Å
C(ar)-C(ar)	1.40Å	1.48Å & 1.44Å
C(ar)-NO ₂	1.47Å	1.42Å & 1.40Å
N-O (nitro)	1.22Å	1.24V
C(sp ²)-C(sp ²)	1.33Å	---
C(sp ²)-N(sp ²)	128Å	---



What are the hazards along the way?

- fires
- chemical burns
- toxic exposures &
- explosions



Risk and Safety Solutions, University of California

