

Learnings from a community safety data sharing experiment

with thoughts on the ethics behind safety data sharing

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“Connecting Professionalism, Safety & Ethics: Opportunities and Challenges “*

LAB SAFETY

University of St. Andrews installs temporary labs to replace those destroyed by fire

Renovations, modular building to cost £12 million

by Dalmeet Singh Chawla, special to C&EN

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“The fire, which broke out on Feb. 10, started when “a routine post-experiment wash-up went wrong,” [the university previously said.](#)”

<https://cen.acs.org/safety/lab-safety/University-St-Andrews-installs-temporary/97/i31>

And again

Dangers of Peroxide Formers—Explosion at UMN

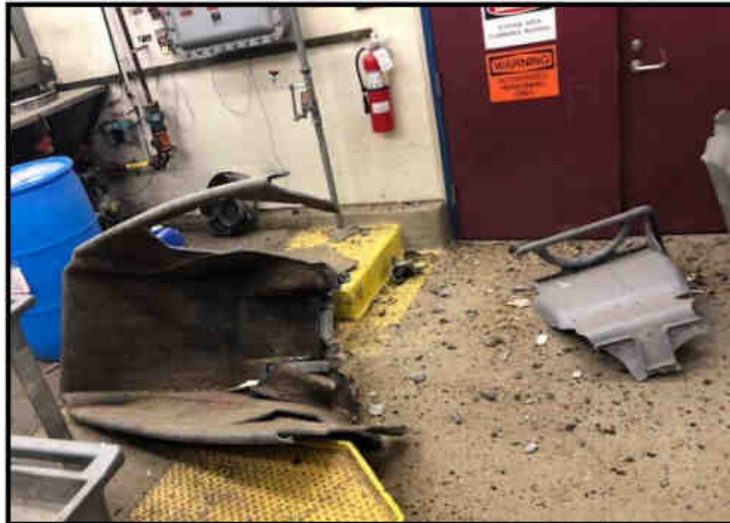


Figure 1: Damaged cart and glass fragmentation from the detonation.

Incident:

A large explosion occurred while a UHS Environmental Health and Safety Technician was processing organic waste from a laboratory cleanout. The technician was in the process of combining flammable solvent waste into a “bulking drum”, which is a standard practice at our facility. Upon disposal of an emptied glass bottle into a tip cart for recycling, a large detonation occurred in the cart. The shockwave from the detonation was large enough to rip the cart in half (Figure 1), blow open the doors in the room, cause damage to adjacent rooms, and shake the windows in the facility. The technician survived the blast without major injuries.

Likely Cause of Explosion:

After a thorough investigation, the most likely cause of the explosion was determined to be **shock sensitive residue from a peroxide forming chemical** deposited inside an empty bottle. While in solution, any shock sensitive crystals may have been stabilized. After the solvent was removed and the residue dried, a detonation of the residual peroxide crystals could have been initiated by friction from the discarded bottles in the cart.

What are Peroxide Forming Chemicals (PFCs)?

PFCs are chemicals that can “auto-oxidize” with atmospheric oxygen under ambient conditions to form organic per-

“The technician was in the process of combining flammable solvent waste into a “bulking drum”, which is a **standard practice** at our facility.”

“The shockwave from the detonation was large enough to rip the cart in half”

https://dehs.umn.edu/sites/dehs.umn.edu/files/safety_alert_-_thompson_explosion_aug19.pdf

How do we do better?

.....“incidents continue to happen. What are we missing?”

Tom Connelly, ACS CEO

Thomas Connelly, The ACS's role in Safety, C&Enews vol 94, iss 26, p. 35
<https://cen-cc-origin.acs.org/articles/94/i26/ACSs-Role-Safety.html>

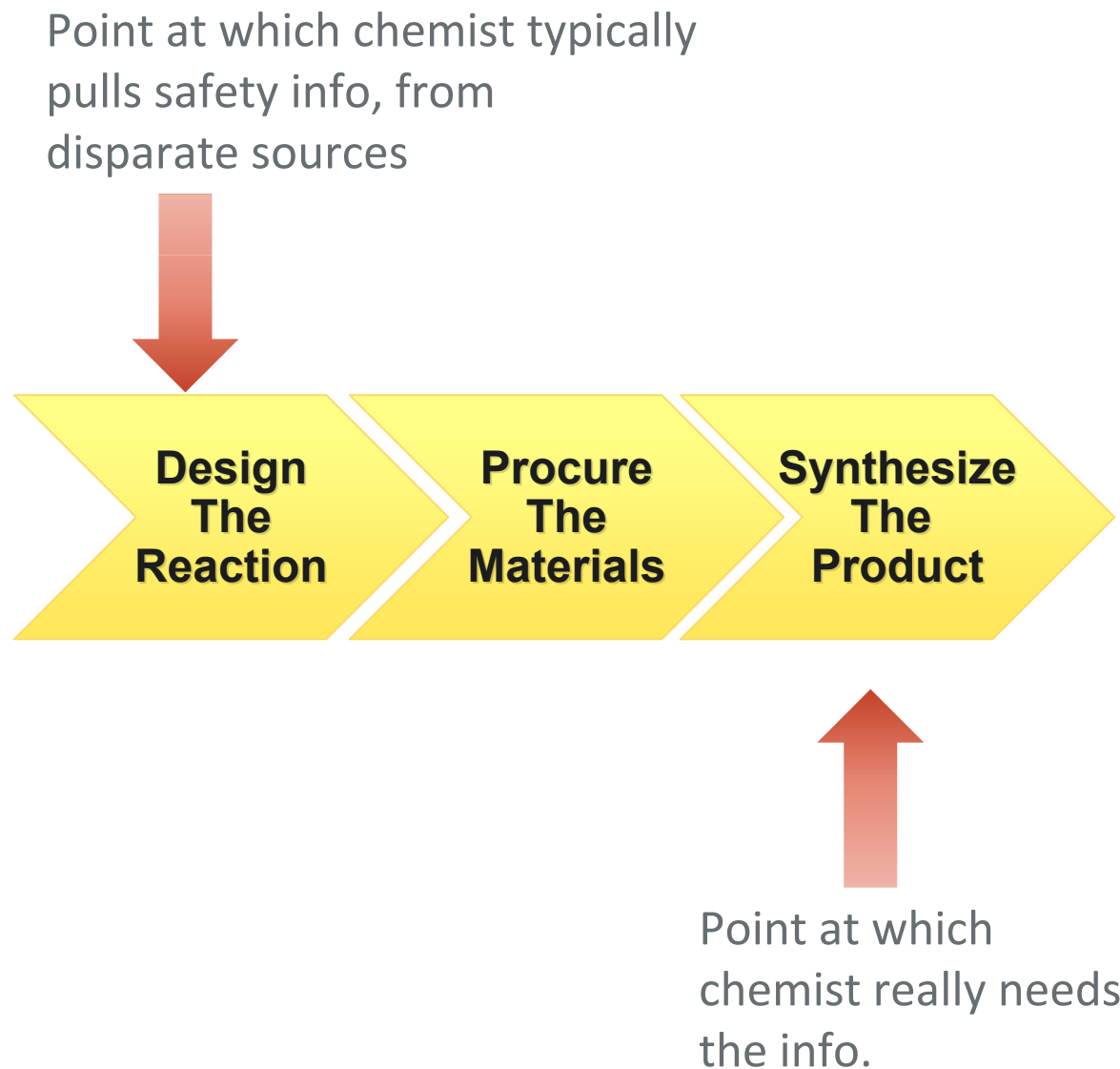


The screenshot shows the top of a C&EN article. The header includes the C&EN logo and navigation links for TOPICS, MAGAZINE, COLLECTIONS, VIDEOS, and JOBS. Below the header, it indicates the article is from Volume 94, Issue 26, page 35, dated June 27, 2016. The article title is "The ACS's Role in Safety" by Thomas Connelly, ACS Executive Director and Chief Executive Officer. A portrait of Tom Connelly is shown on the left. The main text discusses laboratory safety incidents and mentions investigations by the Chemical Safety Board, National Research Council, Association of Public & Land-Grant Universities, and the American Chemical Society. It references a 2012 report from the ACS Committee on Chemical Safety titled "Creating Safety Cultures in Academic Institutions." The article concludes with the question, "Despite these and other guidelines that have been issued, incidents continue to happen. What are we missing?"

What are we missing?

Technical issues

- Data sources can be
 - hard to find
 - scattered all over
 - in unusable formats
 - unknown to the scientist
- Hard to get right data at right time in the right place
- Tools can be difficult to use



What are we missing?

People issues

- Lack of **leadership** in prioritizing safety
- Mistaken perception that safety takes **time** away from important research
- Disinclined to **share** safety learnings
 - Worry, fear & embarrassment
 - IP/liability concerns
 - Inertia
- Blame individuals; forget to look at **systemic factors**
- We are not emphasizing **learning from each other**
- Instilling and cultivating a **culture of safety** is **hard** and **never ends!**



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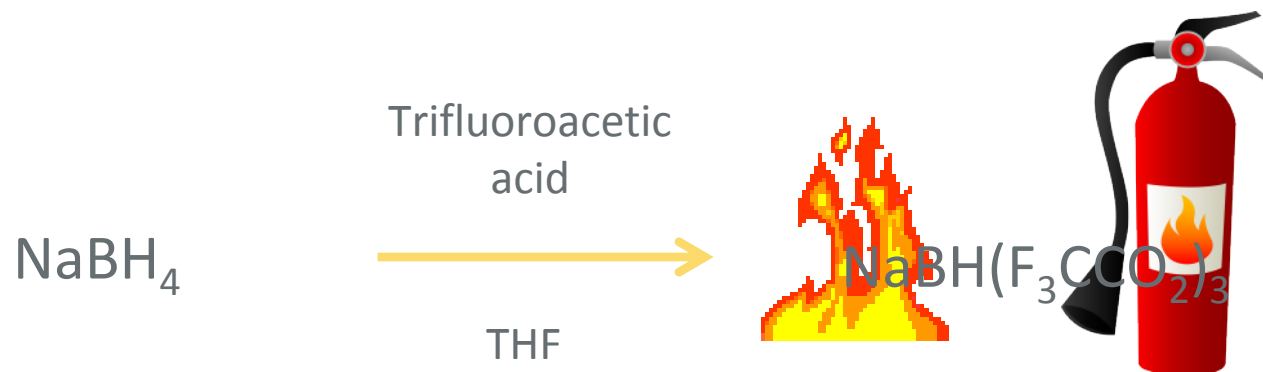
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The Pistoia Alliance project inspiration: Trifluoroacetoxyborohydride

Original synthesis described in United States Patent 4,835,278



- Exothermic Reaction
- Byproduct: hydrogen gas
- Powder dissolves rapidly, reaction not controlled, will consistently catch fire
- Lesson: Must use pelletized NaBH_4 , dissolution controls reaction rate

Ensuring That Lessons Learned Are Not Forgotten, Leveraging ELN to Transform the Safety Paradigm, Mark Manfredi (BMS), ACS Fall 2016 – CHAS Division talk

Why the Pistoia Alliance?

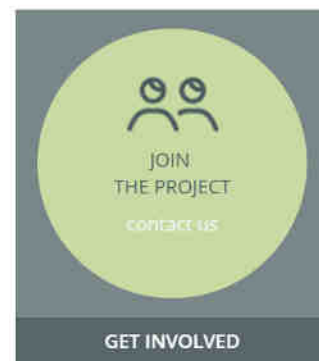
- A global, **not-for-profit** 501(c)(6) alliance of life science companies, technology vendors, publishers, and academics and research institutions.
- Dedicated to **improving life sciences R&D** innovation and effectiveness through **collaborative projects and other activities**
- Provides a proven framework for **cross-industry collaboration**



Goal and value proposition



The Pistoia Alliance Chemical Safety Library project is dedicated to sharing previously inaccessible hazardous reaction information in the interest of increased laboratory and personal safety across the chemical industries.

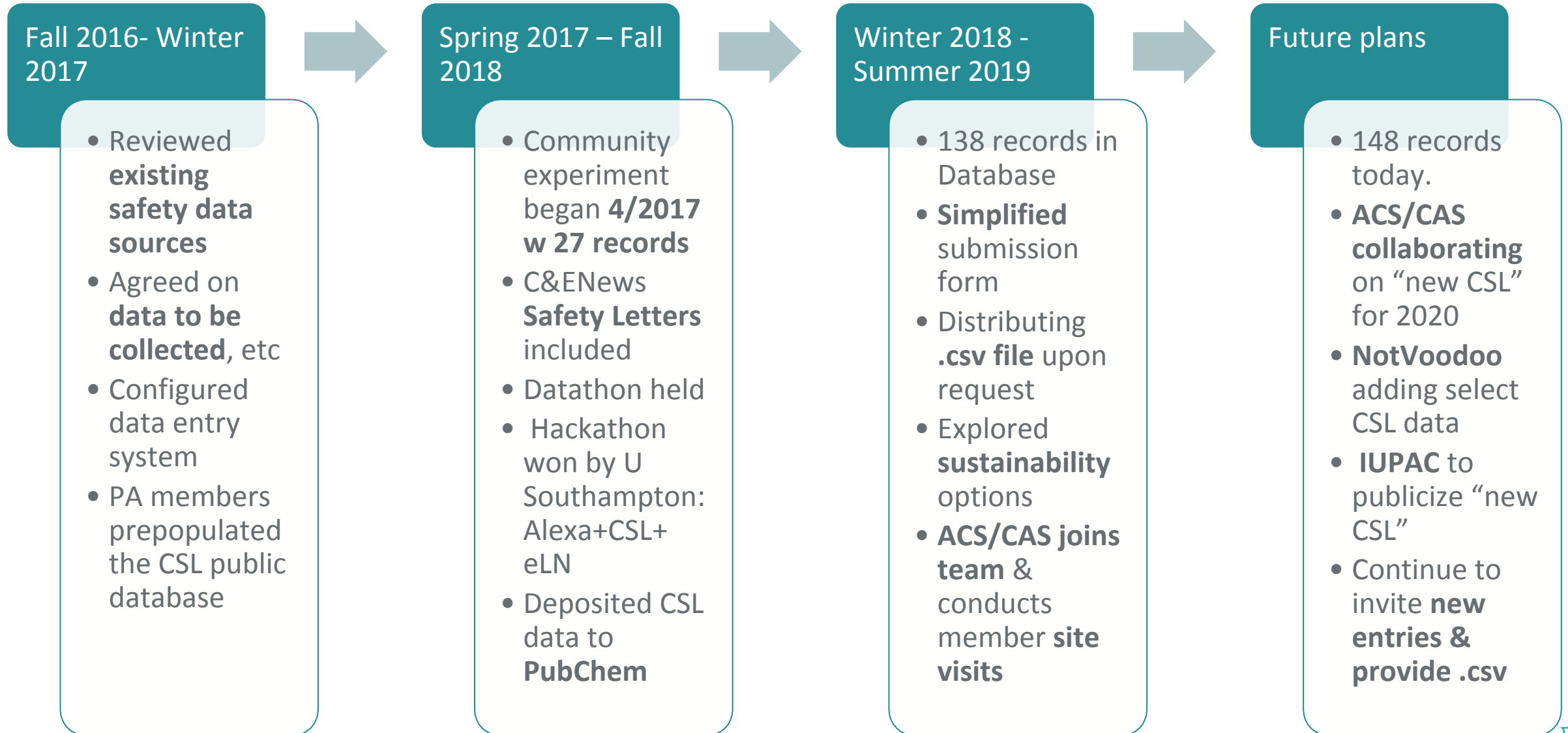


The Pistoia Alliance *Chemical Safety Library* project will **capture** and **share** previously inaccessible **reaction incident** information.

Making this data available to the chemical community at large will allow companies to **learn and avoid reaction incidents** experienced by the wider community, **enhancing overall laboratory safety.**

<https://www.pistoiaalliance.org/projects/chemical-safety-library/>

Chemical Safety Library experiment



Validating the need for actionable safety information

Email requests for the CSL .csv file

Site Type	Year	Requests	Year	Requests	Year	Requests
	2017	154	2018	156	2019	19
Industrial		69		61		11
Academic		65		63		5
Government		12		9		1
Other		8		23		2

Note:

- Publicity efforts throughout 2017 and 2018. No significant publicity run since Fall 2018, due to partner sustainability discussions.
- Recent 2019 uptick tied to renewed public discussion of CSL project, which demonstrates the importance of continued publicity and education.
- Still some hesitation to contribute. Feedback suggests simpler system, clearer anonymity, one-on-one collaboration on data entries, and reminders would increase participation.

A professional imperative to share safety information



Karl Barry Sharpless, W. M. Keck Professor of Chemistry at The Scripps Research Institute (TSRI) and 2019 Priestley Medalist

Thanks to Matt Davenport and Jyllian Kemsley and C&ENews for giving us permission to use this audio excerpt from [Stereo Chemistry: A C&EN Podcast Ep. 20: What happens when you take risks?](https://cen.acs.org/safety/Podcast-Lessons-learned-lab-safety/97/i30) (<https://cen.acs.org/safety/Podcast-Lessons-learned-lab-safety/97/i30>) by [Jyllian Kemsley and Matt Davenport](#) JULY 24, 2019 (Appeared in [VOLUME 97, ISSUE 30](#))

An ethical imperative: why sharing is critical



Ian Tonks, Associate Professor
University of Minnesota

Thanks to Matt Davenport, Jyllian Kemsley, and C&ENews for giving us permission to use this audio excerpt from [Stereo Chemistry: A C&EN Podcast Ep. 20: What happens when you take risks?](https://cen.acs.org/safety/Podcast-Lessons-learned-lab-safety/97/i30) (<https://cen.acs.org/safety/Podcast-Lessons-learned-lab-safety/97/i30>) by [Jyllian Kemsley and Matt Davenport](#) JULY 24, 2019 (Appeared in [VOLUME 97, ISSUE 30](#))

What can you do to promote safety sharing?



- Set up joint safety teams at your University
- Write a safety letter to C&Enews
- To contribute to or get a copy of the CSL contact csladmin@pistoiaalliance.org

Acknowledgements

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