

Putting It All Together



Dmitri Mendeleev

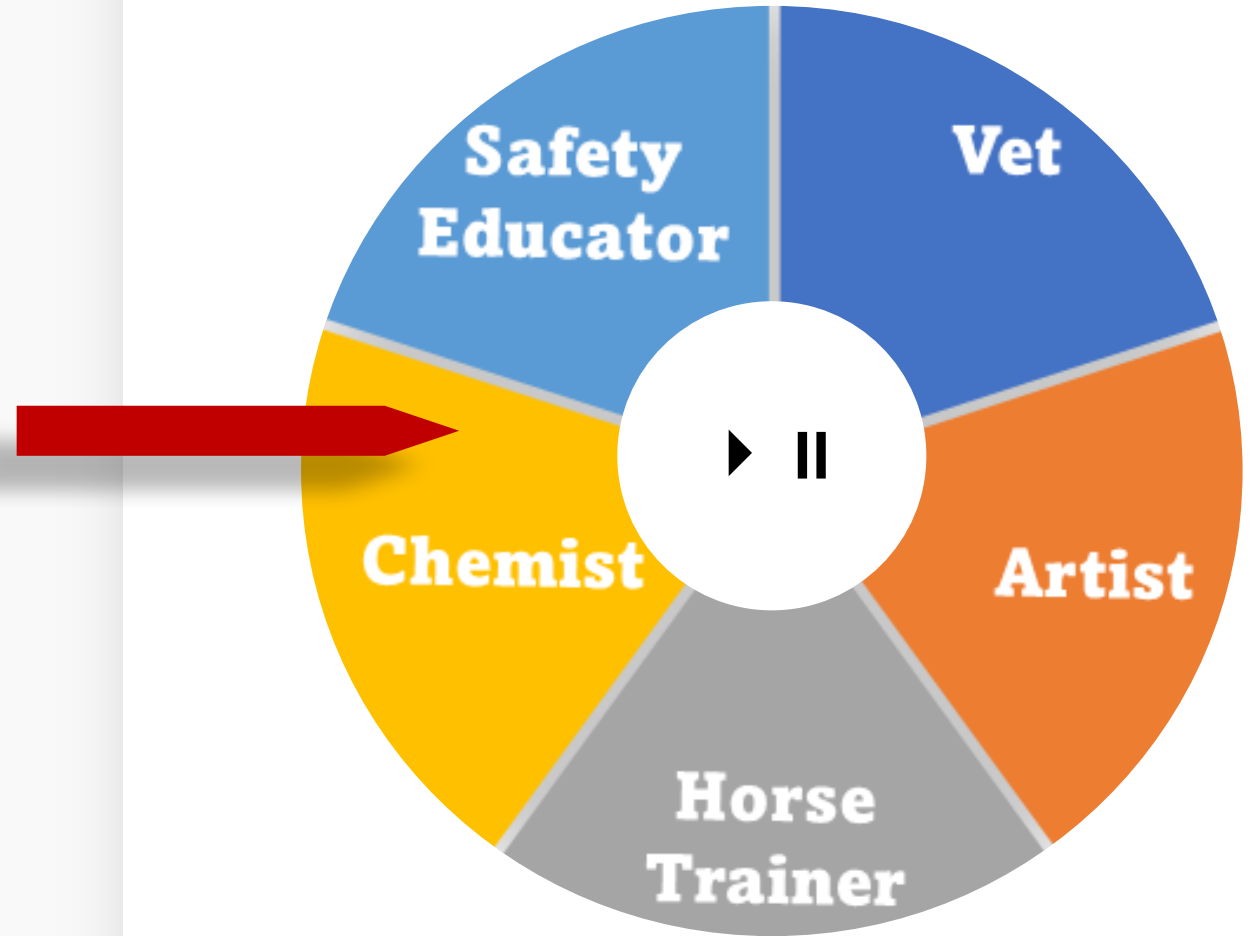
ОПЫТЪ СИСТЕМЫ ЭЛЕМЕНТОВЪ, ОСНОВАННОЙ НА ИХЪ АТОМНОМЪ ВѢСѢ И ХИМИЧЕСКОМЪ СХОДСТВѢ.

			Ti=50	Zr= 90	?=180.
			V=51	Nb= 94	Ta=182.
			Cr=52	Mo= 96	W=186.
			Mn=55	Rh=104,4	Pt=197,1.
			Fe=56	Ru=104,4	Ir=198.
			Ni=Co=59	Pd=106,6	Os=199.
			Cu=63,4	Ag=108	Hg=200.
H=1			Be= 9,4	Mg=24	Zn=65,2
			B=11	Al=27,3	?=68
			C=12	Si=28	?=70
			N=14	P=31	As=75
			O=16	S=32	Se=79,4
			F=19	Cl=35,5	Br=80
					I=127
Li=7	Na=23	K=39	Rb=85,4	Cs=133	Tl=204.
		Ca=40	Sr=87,6	Ba=137	Pb=207.
			?=45	Ce=92	
			?Er=56	La=94	
			?Yt=60	Di=95	
			?In=75,6	Th=118?	

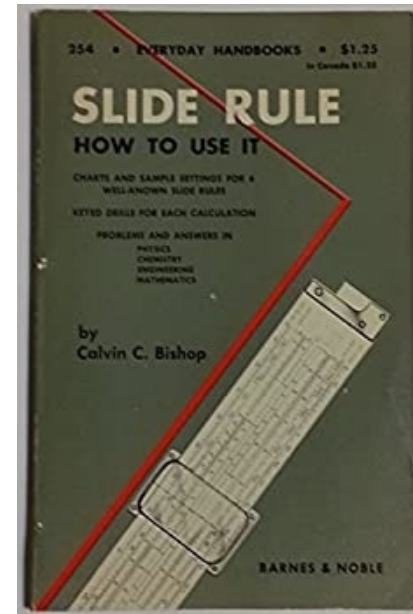
Д. Менделѣевъ

We are all a product of our life experiences. Learning to utilize learned experiences and making choices based on interest and ability took me to unexpected heights.

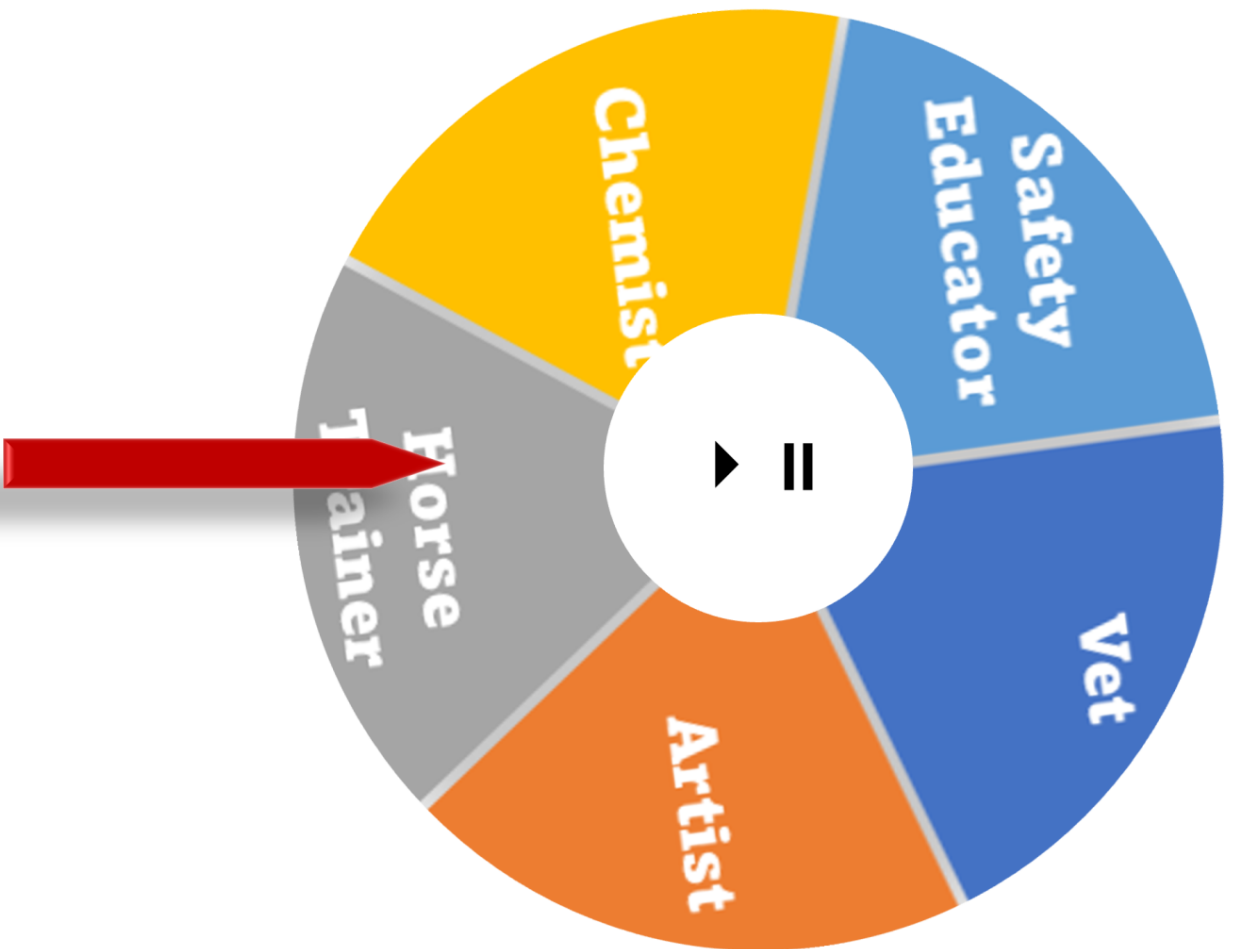
Additionally, utilizing rejections made me resilient.



High School Chemistry



1	D	1	9122	1	11 GPE 1	069	WASHKA	GGYM	BDD
1	D	2	0660	2	2SEM STDY	036	MACADAM	C008	
2	D	3	2134	6	US HIST	091	TARAS	A018	BBBCBBAB
3	D	3	5124	1	SPAN IIA	011	EDELMAN	B009	AABBBBCB
4	D	3	1084	8	PHASE III	032	JOHNSON	A013	CBCCBCBC
5	D	3	4114	3	CHEM B	079	WHITE	B006	CC'E'D'BCD
6	D	3	3074	F	GEOM	053	MUNFAKH	A034	BBCCCCC



Nitrofurazone/DMSO O for Horses



Nitrofurazone is an antibacterial used to treat skin and wounds in horses.



Nitrofurazone/DMSO for Horses

- I learned 40 years ago that DMSO was a carrier agent that enhanced the absorption of other therapeutics.
- I learned to wash my hands BEFORE applying the treatment and to wear gloves.
- I also learned that I am a taster.



Nitrofurazone/DMSO for Horses

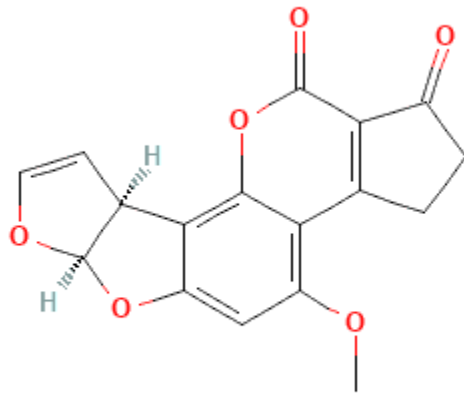
The oral LD₅₀ of dimethyl sulfoxide in the dog is greater than 10 gm/kg.

However, careless use can result in exposure to a toxicant.



Horses are Very Susceptible to Mycotoxins

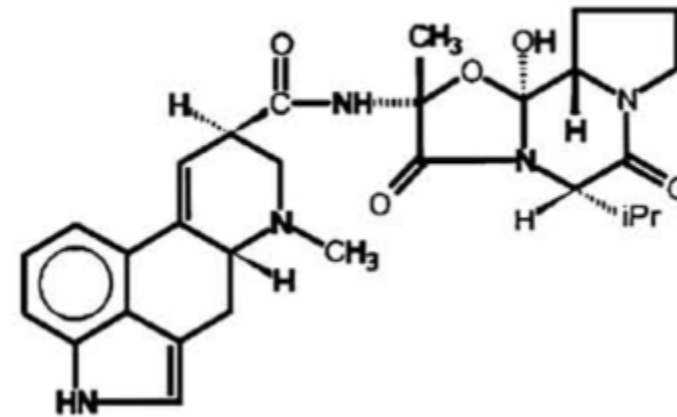
Aflatoxins (*Aspergillus*)



[Aflatoxin B1 PubChem](#)

Fescue Toxicosis

Ergovaline



[R Fayrer-Hosken et al Vol 28, No 11 \(2008\) Journal of Equine Veterinary Science](#)

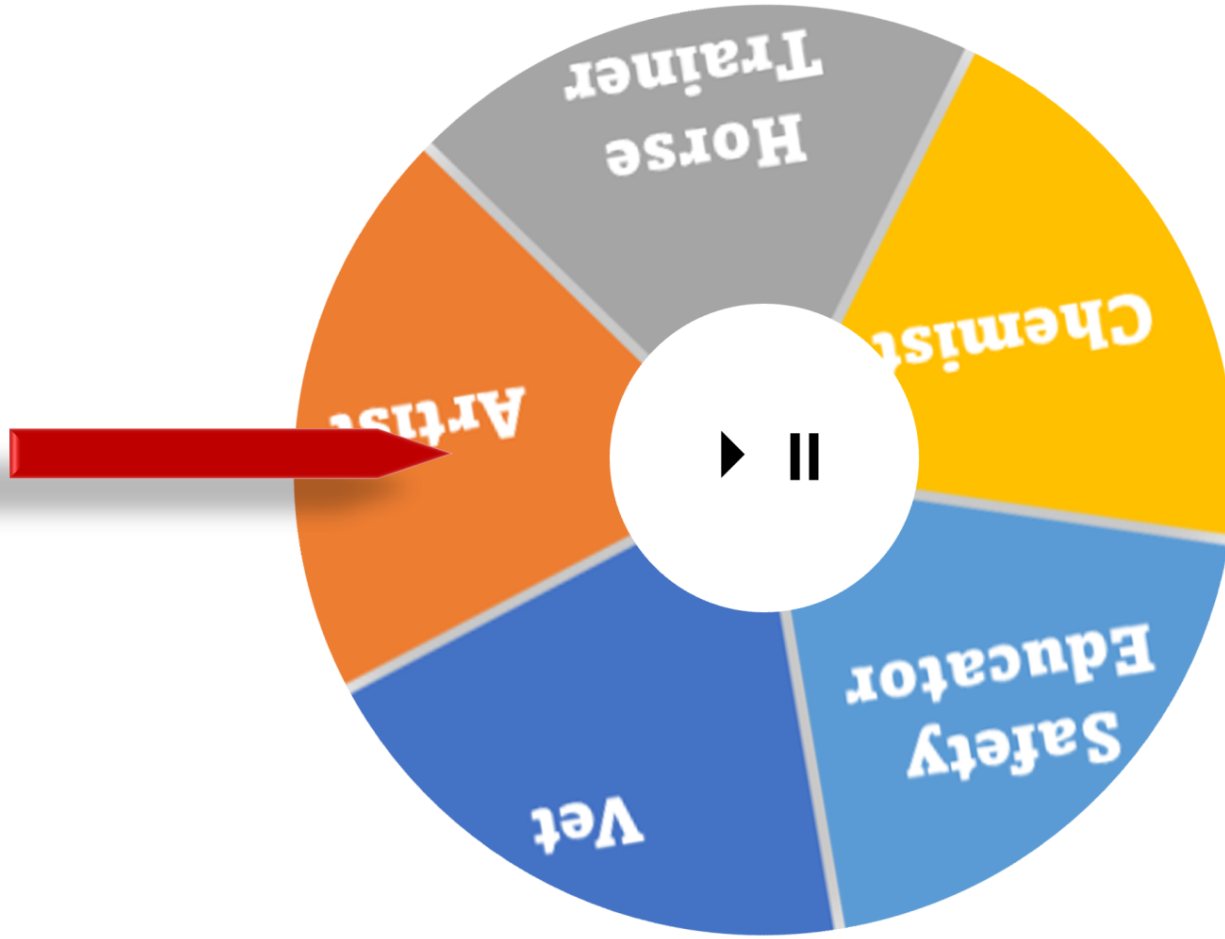
Horseshoeing



I learned that unless it is red, hot metal looks just like cold metal.







Pigments are not just pretty colors



There are a lot of acutely and chronically toxic pigments.

Co	Cr	Ba	Hg	Ar	Cd	Pb
Azurite Cobalt Blue Cobalt Violet Cerulean Smalt	Chromium Oxide Green Terre Verte Viridian	Barium Yellow	Vermilion (Cinnabar)	Realgar Orpiment Emerald green	Cadmium Yellow Cadmium Red	Lead-Tin Yellow Massicot Naples yellow Lead White Minium

<https://www.jcsparks.com/painted/pigment-chem.html#Vermilion>

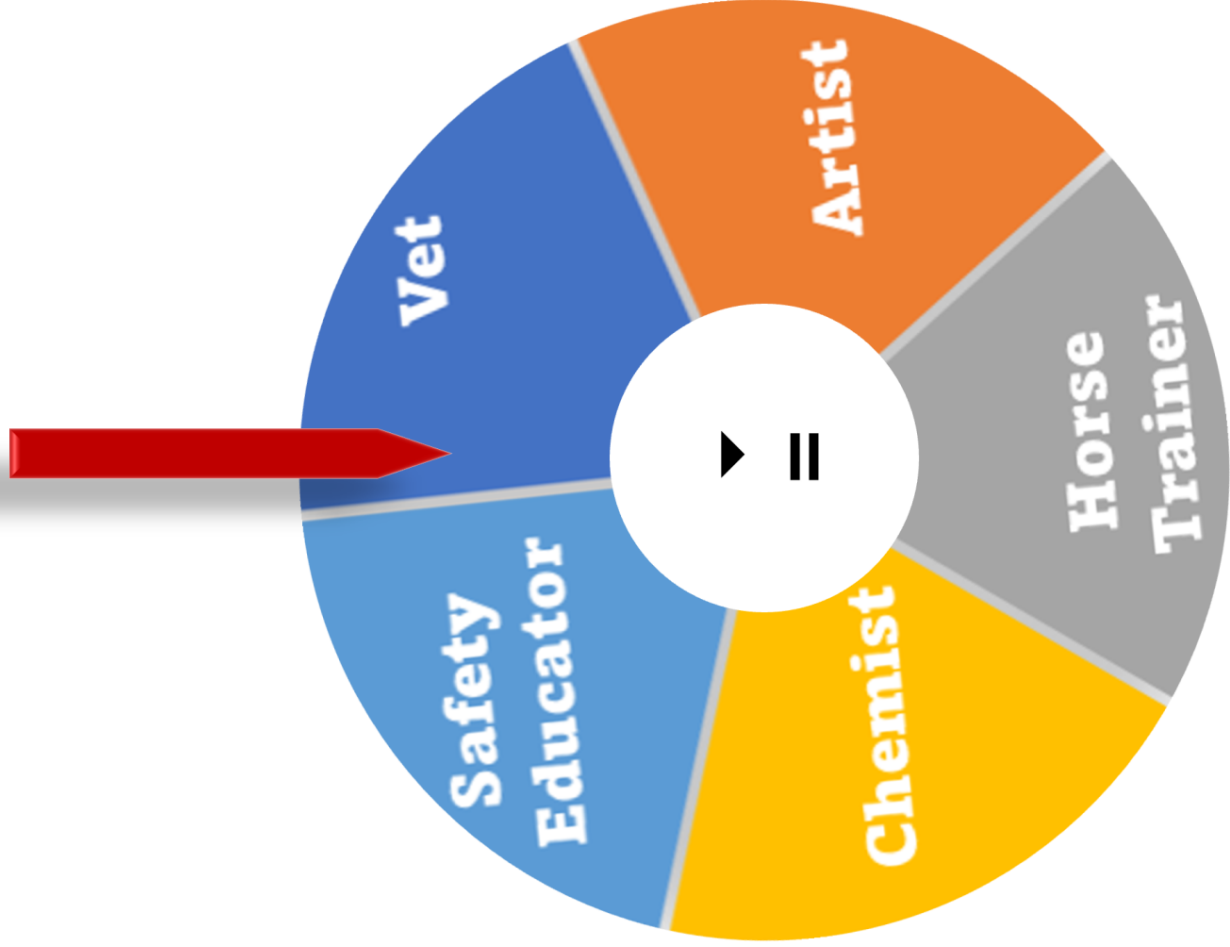
During this time, I learned not “point” your brushes with your mouth...

INORGANIC PAINT PIGMENT COMPOUNDS

A number of inorganic compounds can be used as pigments in paints. Many of these compounds are coloured due to the absorption of light energy by electrons in d orbital subshells, meaning we see colours depending on which wavelengths of light are not absorbed by the compound.

				
CARBON BLACK Carbon, C	CERULEAN BLUE Cobalt (II) stannate, Co_2SnO_4	CHROME GREEN Chromium (III) oxide, Cr_2O_3	COBALT VIOLET Cobalt (II) phosphate, $\text{Co}_3(\text{PO}_4)_2$	CADMIUM ORANGE Cadmium sulfoselenide, Cd_2SSe
				
TITANIUM WHITE Titanium dioxide, TiO_2	ULTRAMARINE BLUE Sulfur-containing sodium silicate, $\text{Na}_4\text{Al}_3\text{Si}_3\text{O}_{10}\text{S}_2$	VIRIDIAN GREEN Hydrated chromium oxide, Cr_2O_3	CADMIUM YELLOW Cadmium sulfide, CdS	CADMIUM RED Cadmium selenide, CdSe
ANTIMONY WHITE Antimony trioxide, Sb_2O_3	PRUSSIAN BLUE Ferric hexacyanoferrate, $\text{Fe}_7(\text{CN})_{18}$	CADMIUM GREEN Cadmium sulfide & chromium (III) oxide	CHROME YELLOW Lead chromate, CdPbO_4	RED OCHRE Iron (III) oxide, Fe_2O_3
ZINC WHITE Zinc oxide, ZnO	COBALT BLUE Cobalt (II) aluminate, CoAl_2O_4		ZINC YELLOW Zinc chromate, ZnCrO_4	



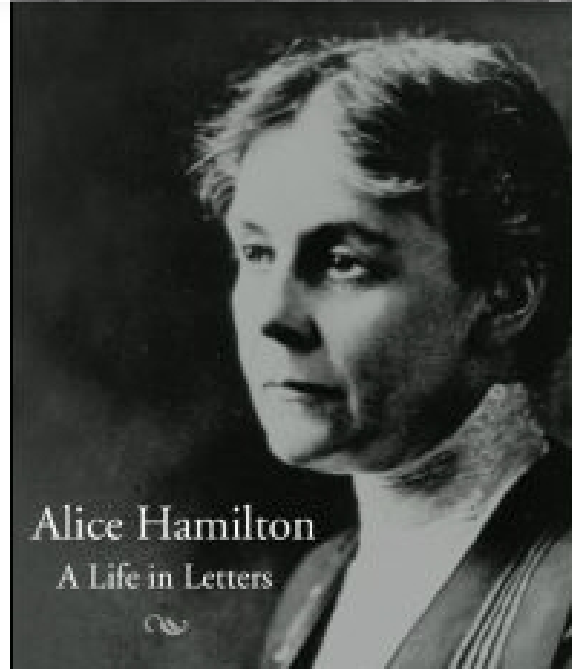


20th Century Advocates

Aldo Leopold (1887 – 1948)

Alice Hamilton (1869 – 1970)

Rachel Carson (1907 – 1964)



Osmium Tetroxide

Pictogram(s)



Corrosive

Acute Toxic

FIXATION OF SUSPENDED TISSUES

A. Preparation of tissue

1. Obtain culture of unicellular organisms. These usually require water heated to 60° C and mixed thoroughly until it dissolves
2. Cool the material until it gels and then cut the material into 1

B. Fixation

1. Transfer tissue to a vial of 3% glutaraldehyde in 0.1 M cacodylate buffer. After one hour, transfer material to refrigerator at 4° C.
2. Tissue may be processed immediately or held up to several hours in a fixative solution of 0.1 M cacodylate buffer at pH 7.4 and 4° C.
3. Rinse tissue with three changes of cold buffer, 10 minutes per change.
4. Postfix in cold 2% osmium tetroxide (mixed 1:1 4% osmium with the 2X buffer) for 30 minutes to 2 hours on ice or in the refrigerator. Work with uncapped osmium solutions only under the fume hood.
5. Rinse tissue with cold buffer for 10 minutes.

C. Dehydration

1. Dehydrate the tissue using the following concentrations of ethanol in a cold graded ethanol series (10 minutes per change): 30%, 50%, 70%, 80%, 90%, 100%. If you have to stop at any point in the dehydration process 70% ethanol is best.
2. Continue dehydration with two more changes of cold 100% ethanol (10 minutes per change).
3. Follow cold 100% ethanol with one 10 minute change of cold (1:1) 100% ethanol : propylene oxide. Use propylene oxide in the fume hood.
4. Complete dehydration with three changes of propylene oxide in the refrigerator, 10 minutes per change. Bring the tissue slowly to room temperature.

D. Infiltration

1. Infiltrate tissue using the following concentrations of Spurr's resin, approximately 2-4 hours per change or longer: (1:2) Spurr's resin : prop. oxide, (2:1) Spurr's resin : prop. oxide. Refractory material may take up to one day per change.
2. Use two changes of 100% Spurr's resin (2 hours or more per change).
3. Embed in fresh Spurr's resin (less than one day old). Polymerize for 6 to 8 hours at 70° C.

Signal

Danger

GHS Hazard Statements

H300: Fatal if swallowed [Danger Acute toxicity, oral]

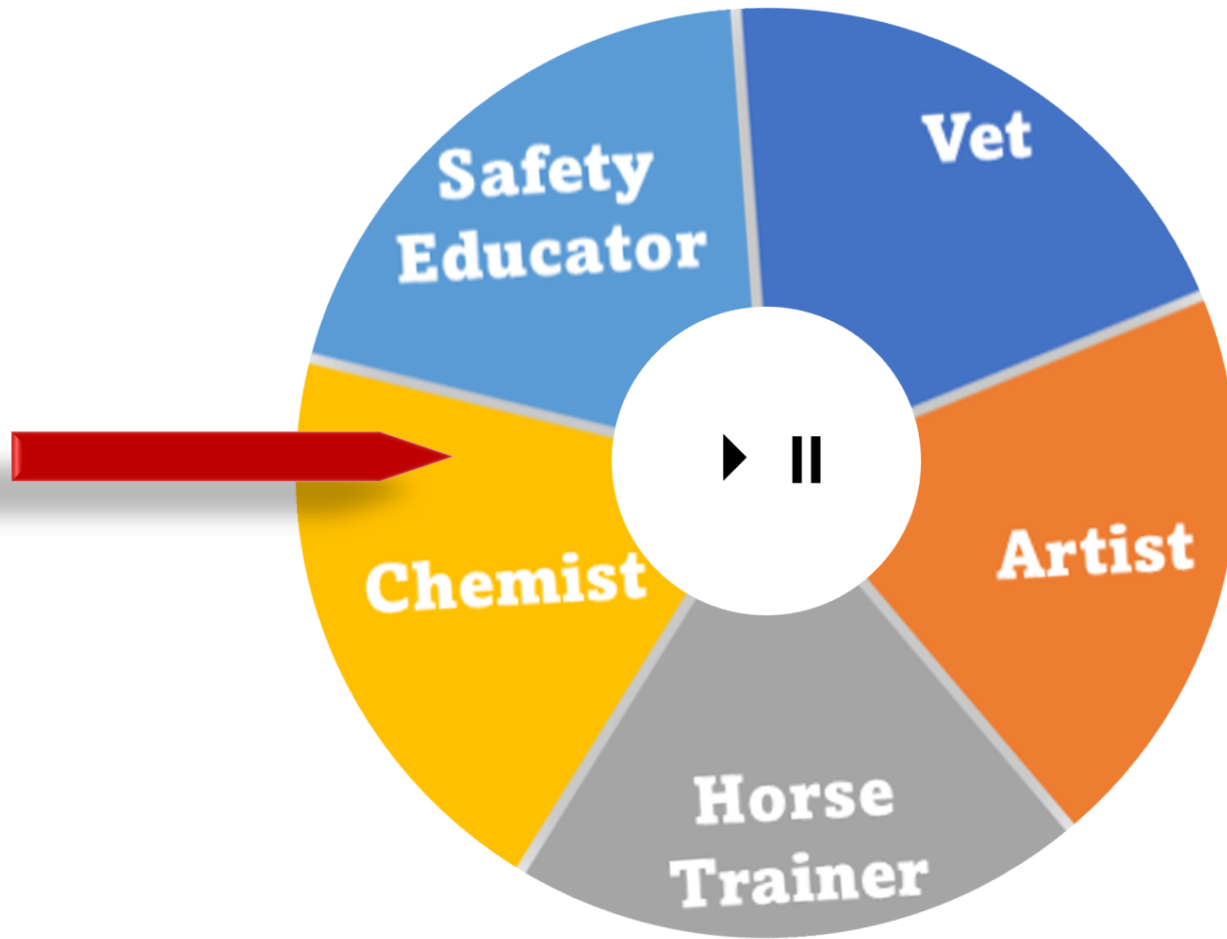
distilled

H310: Fatal in contact with skin [Danger Acute toxicity, dermal]

H314: Causes severe skin burns and eye damage [Danger Skin corrosion/irritation]

H330: Fatal if inhaled [Danger Acute toxicity, inhalation]

t results!



Chemistry

They have calculators
now!! Yeah.





Reactivity of Nitric Acid

All of my graduate research required using 1+1 HNO_3 .

Our AAS lab sink had glass traps and my research advisor warned that any combination of nitric acid and acetone would likely blow the glass trap.

I did not even allow acetone or other solvents in the lab.





All that is gold does not glitter,

Not all those who wander are lost;

The old that is strong does not wither,

Deep roots are not reached by the frost.

From the ashes a fire shall be woken,

A light from the shadows shall spring;

*Renewed shall be blade that was
broken,*

The crownless again shall be king.

— J.R.R. Tolkien, *The Fellowship of the Ring*

So, Where Was I?

FROM HORSES

- I learned chemicals could be absorbed.
- I learned that hazards can be isolated to minimize risk and that routines (SOPS) minimize risk.
- I learned that there are many plant toxins that affect horses and humans.

FROM ART

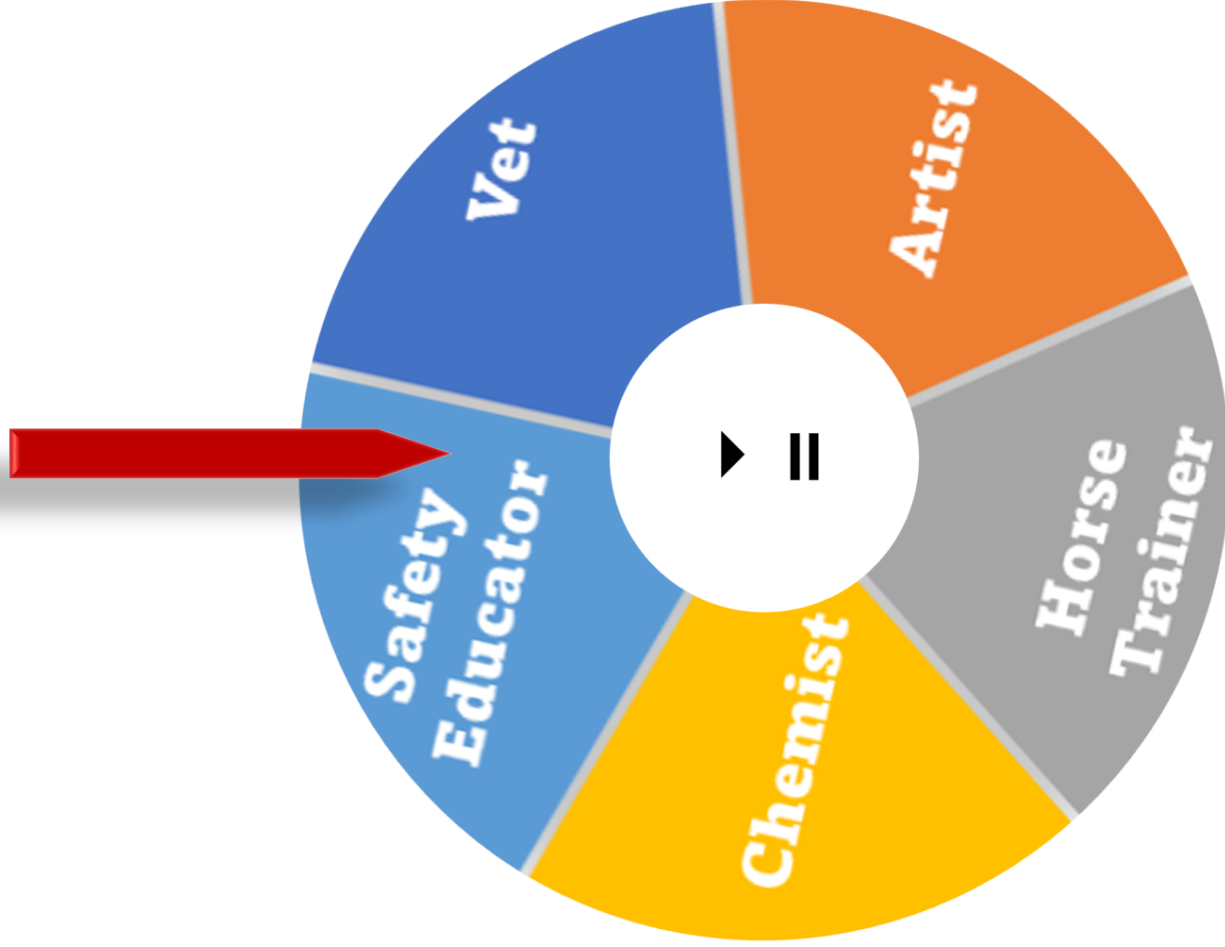
- I learned that toxicants can be ingested.
- I learned that solvents cause dermatitis and dissolve gloves.

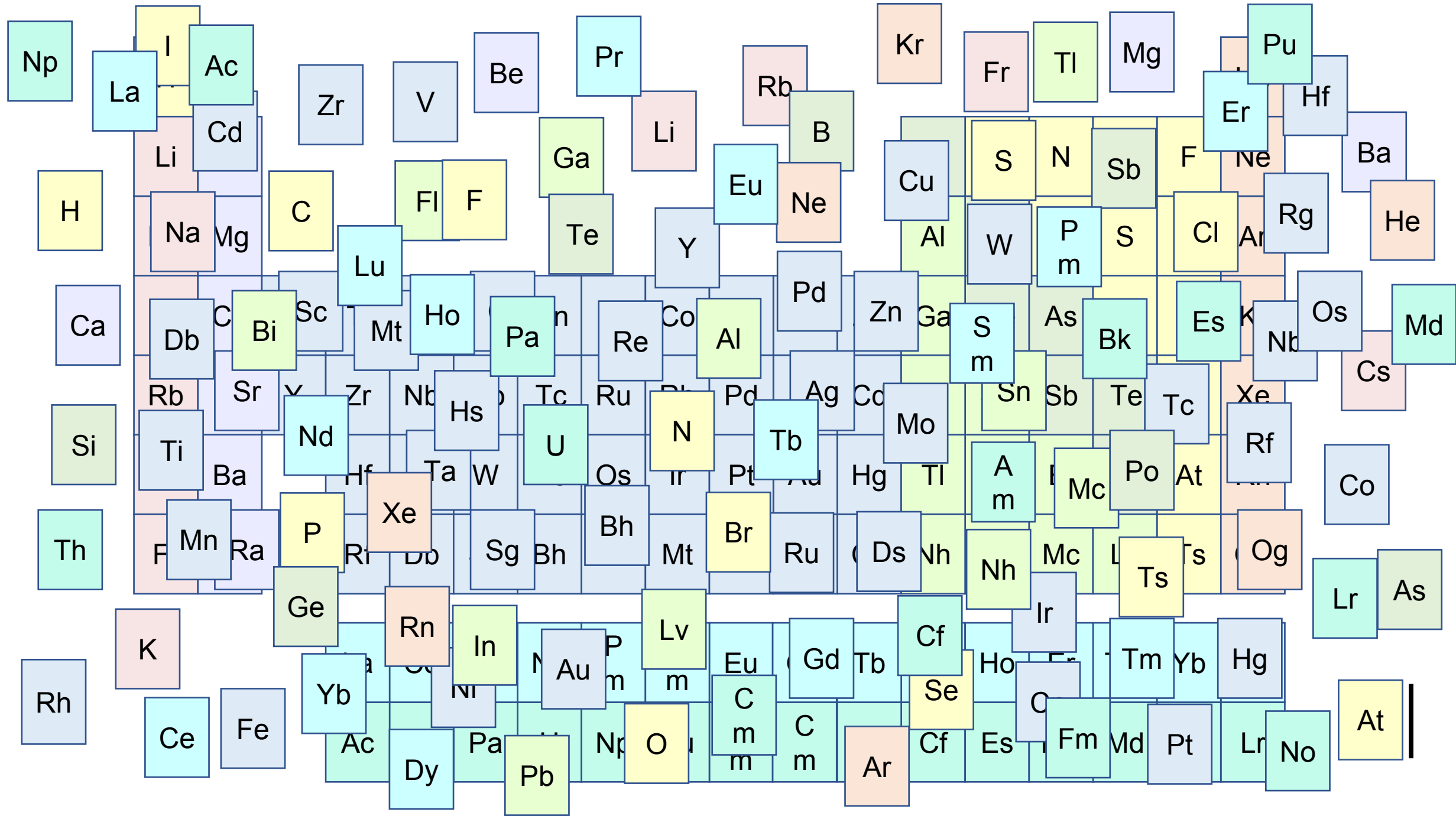
FROM BIOLOGY

- I learned about how toxins affect systems.
- I learned how toxicants affect reproductive development.
- I learned that osmium tetroxide is nasty stuff....
- I learned about environmental pollutants, responsible use of chemicals, bioaccumulation, and biomagnification.
- I learned about inhalation hazards ventilation in labs.

FROM CHEMISTRY

- I learned not to mix nitric acid with acetone – and a lot of other stuff....





Lifelong Leadership Skills

Work hard

Be resilient

Adapt

Know your own worth

Say “yes”

Do a lot of things until you find out what you are good at because everything you learn will make you better at that job...

Don't be afraid to
spin the wheel.

Thank you!

